Cosmic-ray exposure ages of enstatite chondrites

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Abstract: Eleven enstatite (E-) chondrites were analyzed for abundances and isotopic ratios of noble gases by total melting. Light noble gases, He, Ne and Ar are discussed. We obtained cosmic-ray exposure ages (T^{21}) based on cosmogenic ²¹Ne that are divided into two groups, below 15 Myr and above 40 Myr. Both ranges of T^{21} contain various petrologic-types of E-chondrites. Isotopic ratios and concentration of Ne for Yamato (Y)-8414 and -86004 are almost the same, which suggests that they are from a common fragment. Identical exposure age T^{21} of 10 Myr for Y-8414 and -86004 supports their pairing. The pairing for Y-8414 and -86004 based on noble-gas signatures and T^{21} is consistent with their petrologic similarities. Pairing for Y-792959 and -793161 is also suggested by their similarities in mineral compositions, their close sampling location and the same petrologic-type. However, there are discrepancies in He and Ar contents between the two samples, which fails to support their pairing.

Including literature data, the exposure ages show no systematic correlation with petrologic type. Additionally, both EH- and EL-chondrites are found in both two peaks of T^{21} . The exposure age distribution has no implications about the structure of E-chondrite parent body. The parent body of E-chondrites has possibly lost the original structure after many collisions in the early solar system evolution.

1. Introduction

Enstatite (E) chondrites are assemblages of highly reduced minerals. They are divided into two subclasses based on their Fe contents in bulk stones: high-Fe (EH) and low-Fe (EL) groups (Sears *et al.*, 1982; Kallemeyn and Wasson, 1986). There is no continuity in their bulk compositions of nonvolatile major elements between the two groups (Keil, 1968; Baedecker and Wasson, 1975) and the hiatus in this chemical composition could not be accounted for by simple planetary metamorphism (Keil, 1968). In this context, separate parent bodies for EH- and EL-chondrites have been proposed by some authors (*e.g.*, Baedecker and Wasson, 1975; Sears *et al.*, 1982; Keil, 1989; Zhang *et al.*, 1995).

On the other hand, Wasson and Wai (1970) proposed a single parent body common to EH- and EL-chondrites based on the continuous variation of mineral composition. Biswas *et al.* (1980) and Kaczaral *et al.* (1988) also suggested a

single parent body from volatile/mobile trace element trends. Oxygen isotopic compositions of the two subclasses that occupy the narrow region along the terrestrial fractionation line support these models on the single E-chondrite parent body (Clayton *et al.*, 1984). Recently, Kong *et al.* (1997) suggested a single parent body and two-stage metamorphism for EH- and EL-chondrites based on continuity in composition of moderately volatile elements, and on inverse variation of the abundances of these elements with petrologic type. In addition, Akridge and Sears (1999) suggested that the fractionation between metal and silicate can be produced on a single parent body. So the difference in bulk Fe-contents between EH- and EL-chondrites is not strong evidence for the different parent bodies for these chondrites.

Some noble-gas signatures characteristic of the E-chondrites have been reported: (1) high ³⁶Ar/¹³²Xe ratio compared to other chondrite classes (Zähringer, 1962, 1968; Crabb and Anders, 1981; Wacker and Marti, 1983), (2) absence of extensive correlation between trapped noble gas concentrations and petrologic type (Zähringer, 1966; Marti, 1967), and (3) high ¹²⁹Xe/¹³²Xe ratio relative to other chondrite classes (Reynolds, 1960; Zähringer, 1962; Crabb and Anders, 1981, 1982; Wacker and Marti, 1983). In addition, cosmic-ray exposure ages for eighteen E-chondrites showed a systematic difference with respect to petrologic type (Crabb and Anders, 1981): E4 chondrites have exposure ages below 16 Myr and most of E6's above 30 Myr. Based on this observation, Crabb and Anders (1981) suggested that the parent body has not been extensively mixed and accordingly a single petrologic-type dominates on the about 1 km scale of individual impacts.

Cosmogenic noble gases are produced by interactions between nuclides in meteoritic materials and galactic cosmic-rays. Thus, cosmic-ray exposure ages reflect recent irradiation records. An aim of this study is to examine the correlation between the exposure age distribution and petrologic types for E-chondrites. Our samples include EH6 chondrites which have not been measured by Crabb and Anders (1981). Pairing of the chondrites are also mentioned based on noble gas compositions. In this paper, light noble gases are discussed although all noble gases were measured. Full data sets of Kr and Xe, and discussion on their isotopic compositions will be presented elsewhere with data sets obtained by stepped crushing and heating experiments.

2. Samples and experimental methods

Samples are nine Antarctic enstatite chondrites (Allan Hills (ALH)-77295, Yamato (Y)-691, -74370, -791790, -792959, -793161, -793225, -8414, and -86004) and two non-Antarctic ones (Ilafegh 009 and Pillistfer), as listed in Table 1. They are eight EH- and two EL-chondrites, while Y-793225 is classified as petrologic-type 6, intermediate between EH- and EL-chondrites on the basis of the composition of (Mg, Mn, Fe)S solid solution (Lin and Kimura, 1998).

Noble gas analyses were performed by a conventional total melting method in order to reduce experimental uncertainties in concentrations and isotopic ratios. Samples were sliced in about 1 mm thickness using ethanol as a lubricant, and wrapped in Al-foil. They were mounted in a sample holder of an extraction furnace, and preheated at 150° C for one day so as to desorb atmospheric gases. Noble gases were extracted by heating the samples at 1800°C for 20 min. Evolved reactive gases (e.g. CO_2 , H_2O and N_2) were eliminated using two hot Ti-Zr and two SORB-AC getters. Purified noble gases were separated into four fractions (He-Ne, Ar, Kr, and Xe) with two charcoal traps and a temperature-controlled cryogenic trap. Noble gases were measured with a modified VG5400 mass spectrometer (MS-II) at Laboratory for Earthquake Chemistry, University of Tokyo (e.g., Nagao et al., 1999). Sensitivities and mass discrimination of the mass spectrometer were calibrated by measuring known amounts of atmospheric noble gases and an isotopic ratio of helium standard gas (${}^{3}\text{He}/{}^{4}\text{He}=1.71\times10^{-4}$) prepared by mixing of known amounts of ³He and ⁴He. The typical sensitivities (S) and correction factors (M) for the mass discrimination were as follows: $S(^{4}He) = 3 \times 10^{7}$, $S(^{20}Ne) = 5 \times 10^{7}$ and $S(^{36}Ar)$ $=1 \times 10^8$ in volts/cm³ STP; M(³He/⁴He) = 0.99 \pm 0.01, M(²⁰Ne/²²Ne) = 0.955 ± 0.001 , M(²¹Ne/²²Ne) = 0.978 ± 0.002 , M(³⁸Ar/³⁶Ar) = 1.018 ± 0.001 and $M(^{40}Ar/^{36}Ar) = 1.036 \pm 0.002$. To check the calibration for sensitivities and mass discrimination, ALH-769 (L6), a laboratory standard sample, was measured in the series of total melting analyses. Blanks for noble gases at 1800°C are ${}^{4}\text{He}=6\times$ 10^{-10} , ${}^{20}Ne=2\times10^{-12}$, ${}^{36}Ar=5\times10^{-12}$ in cm³ STP, and their isotopic ratios are atmospheric but slightly affected by memory from previous measurements of cosmogenic gases, except for a ⁴⁰Ar/³⁶Ar ratio. Errors sited in Table 1 are statistical ones (1 σ). Uncertainties for the noble gas concentrations are estimated to be about 10%.

Bulk chemical compositions for Y-74370, -791790 and -793225 were also determined by electron probe microanalyzers (JEOL 733 superprobe and JEOL XA-8800R). The analyses were conducted at 12 nA and 15 kV with electron beam of 50 μ m in diameter. The results are shown in Table 3. Since the analyses were made by assuming all cations to be oxides, oxygen contents in each sample are calculated to total up to 100 wt%.

3. Results and discussion

3.1. Noble gas concentrations and isotopic ratios

Table 1 shows noble gas isotopic ratios and concentrations of He, Ne and Ar. Analyzed samples weigh 5.0–19.1 mg (Table 1). Except for ALH-77295, ³He in samples is cosmogenic, while ⁴He is dominantly radiogenic. As shown in Ne isotope plot (Fig. 1a), ALH-77295 contains solar Ne, which has already been reported by others (Signer *et al.*, 1983; Wieler *et al.*, 1985). Solar He and Ar should be contained in this meteorite. Figure 1 also shows that Ne in equilibrated Echondrites is dominated by a cosmogenic component, while Ne in unequilibrated ones is generally a mixture of cosmogenic and trapped ones. Isotopic ratios of Ar indicate that all samples contain cosmogenic and radiogenic components in addition to trapped one.

| Sample | Class ³⁾ | Analyzed weight (mg) | ³ He | ⁴ He | ³He∕⁴He | ²⁰ Ne | ²¹ Ne | ²² Ne | ²⁰ Ne/ ²² Ne | ²¹ No/ ²² No | ³⁶ Ar | ³⁸ Ar | 40Ar | ³⁸ Ar/ ³⁶ Ar | ⁴⁰ Ar/ ³⁶ Ar |
|-------------|---------------------|----------------------------|-----------------|-----------------|---------------------|------------------|------------------|------------------|------------------------------------|------------------------------------|------------------|------------------|-------|------------------------------------|------------------------------------|
| Y-691 | EH3 | 16.9 | 29.7 | 4750 | 0.006256 ± 0.000046 | 28.7 | 9.01 | 12.9 | 2.2211 ± 0.0027 | 0.6983 ± 0.0016 | 111 | 21.5 | 12600 | 0.19361 ± 0.00028 | 113.34 ± 0.26 |
| Y-791790 | EH3 | 11.0 | 35.5 | 7920 | 0.004480 ± 0.000038 | 44.0 | 11. 2 | 15.6 | 2.8206 ± 0.0070 | 0.7191 ± 0.0019 | 1480 | 278 | 67600 | 0.18812 ± 0.00016 | 45.678 ± 0.047 |
| Y-792959 | EH3 | 10.5 | 485 | 10200 | 0.04752 ± 0.00044 | 97.6 | 92.2 | 109 | 0.8956 ± 0.0016 | 0.8456 ± 0.0064 | 55.0 | 17.8 | 29300 | 0.32342 ± 0.00077 | 533.0 ± 1.1 |
| Y-793161 | E3 | 11.1 | 247 | 5500 | 0.04498 ± 0.00037 | 85.7 | 77.5 | 91.2 | 0.9399 ± 0.0020 | 0.8493 ± 0.0013 | 87.1 | 23.0 | 55400 | 0.26407 ± 0.00084 | 636.1 ± 1.3 |
| ALH-77295 | EH4 | 8.7 | 694 | 505000 | 0.001374 ± 0.000010 | 2140 | 143 | 315 | 6.786 ± 0.012 | 0.4542 ± 0.0018 | 269 | 60.2 | 52800 | 0.2239 ± 0.0011 | 196.13 ± 0.46 |
| Y-74370(A) | EH4 | 10.5 | 10.6 | 2550 | 0.004138 ± 0.000026 | 13.9 | 4.23 | 5.93 | 2.344 ± 0.011 | 0.7131 ± 0.0036 | 26.2 | 5.08 | 25722 | 0.19402 ± 0.00055 | 981.8 ± 2.0 |
| Y-74370(B) | EH4 | 19.1 | 24.0 | 6500 | 0.003685 ± 0.000030 | 18.4 | 5.60 | 7.80 | 2.3548 ± 0.0097 | 0.7184 ± 0.0025 | 67.1 | 12.9 | 55591 | 0.19298 ± 0.00019 | 828.5 ± 1.4 |
| Y-8414 | EH6 | 7.2 | 134 | 8960 | 0.01499 ± 0.00015 | 21.8 | 20.4 | 25.4 | 0.8595 ± 0.0046 | 0.8040 ± 0.0040 | 13.0 | 5.15 | 65500 | 0.3949 ± 0.0030 | 5022 ± 101 |
| Y-86004 | EH6 | 5.0 | 101 | 13200 | 0.00766 ± 0.00011 | 18.8 | 18.0 | 22.8 | 0.8242 ± 0.0078 | 0.7917 ± 0.0033 | 7.50 | 3.82 | 46300 | 0.5100 ± 0.0057 | 6178 ± 137 |
| Y-793225 | EH/EL6 | 7.4 | 562 | 8280 | 0.06790 ± 0.00022 | 99.8 | 101 | 121 | 0.8263 ± 0.0013 | 0.8399 ± 0.0032 | 469 | 96.1 | 38200 | 0.20506 ± 0.00041 | 81.564 ± 0.053 |
| Pillistfer | EL6 | 12.7 | 76.8 | 10700 | 0.007180 ± 0.000053 | 29.3 | 28.2 | 30.8 | 0.9503 ± 0.0013 | 0.9168 ± 0.0013 | 289 | 56.9 | 67200 | 0.19686 ± 0.00024 | 232.47 ± 0.46 |
| llafegh 009 | EL-AN | 13.3 | 128 | 8960 | 0.014236 ± 0.000040 | 38.6 | 38.4 | 42.4 | 0.9089 ± 0.0021 | 0.9055 ± 0.0037 | 41.3 | 11.7 | 38000 | 0.28300 ± 0.00091 | 921.60 ± 2.20 |

Table 1. Isotopic ratios¹⁾ and concentrations²⁾ of He, Ne and Ar.

Statistical 1σ errors are given for isotopic ratios.
Concentrations of He, Ne and Ar are in 10⁻⁹ cm³ STP/g, and their uncertainties are estimated to be about 10% for 1σ.
References for classification are summarized in Appendix.



Fig. 1. Ne-three isotope diagram. Literature data for ALH-77295 (Wieler et al., 1985), Y-691 (Shima et al., 1973) and Pillistfer (Crabb and Anders, 1981) are also plotted for comparison. (a): Some EH-chondrites show existence of trapped Ne as well as cosmogenic one. (b): Enlarged cosmogenic area in the lower right of Fig. 1a. The least square fitting line was used to obtain cosmogenic ²¹Ne/²²Ne ratios (see text).

| Sample | Class | [³ He]c ²⁾ | [²¹ Ne]c | [³⁸ Ar]c | $\left(\frac{21}{20}\right)$ | F ³ | F ²¹ | F ³⁸ | P ³ | P ²¹ | P ³⁸ | T3 | T ²¹ | T ³⁸ |
|-------------|-----------------------|-----------------------------------|------------------------------------|----------------------|-------------------------------------|----------------|-----------------|-----------------|--------------------|---------------------|-----------------|------------|-----------------|-----------------|
| | | (×1 | 0 ⁻⁹ cm ³ ST | P/g) | (^{2 2} N •) ₀ | | | - | (×10 ⁻⁹ | cm ³ STP | /g/Myr) | | (Myr) | |
| Y-691 | EH3 | 30 | 9.0 | 0.71 | 0.82 | 1.00 | 0.85 | 0.87 | 16 | 1.9 | 0.34 | 1.9 | 4.8 | 2.1 |
| Y-791790 | EH3 | 36 | 11 | 0.20 | 0.91 | 1.00 | 0.97 | 0.65 | 16 | 3.4 | 0.30 | 2.2 | 3 .3 | 0.7 |
| Y-792959 | EH3 | 490 | 92 | 8.5 | 0.846 | 1.00 | 0.78 | 1.06 | 16 | 2.0 | 0.44 | 31 | 47 | 20 |
| Y-793161 | E3 | 250 | 78 | 7.6 | 0.849 | 1.00 | 0.77 | 0.88 | 16 | 2.0 | 0.36 | 16 | 40 | 21 |
| ALH-77295 | EH4 | 500 | 140 | 11 | 0.91 | 1.00 | 0.81 | 0.98 | 16 | 2.9 | 0.46 | 31 | 48 | 24 |
| Y-74370 | EH4 | 24 | 5.6 | 0.38 | 0.85 | 1.00 | 0.82 | 0.78 | 16 | 2.0 | 0.32 | 1.5 | <i>2.</i> 7 | 1.2 |
| Y-8414 | EH6 | 130 | 20 | 3.1 | 0.804 | 1.00 | 1.00 | 0.60 | 16 | 2.1 | 0.22 | 8.4 | 10 | 14 |
| Y-86004 | EH6 | 100 | 18 | 2.8 | 0.792 | 1.00 | 0.88 | 0.69 | 15 | 1.7 | 0.24 | 6.5 | 10 | 11 |
| Y-793225 | EH/EL ³⁾ 6 | 560 | 100 | 9.7 | 0.840 | 1.00 | 0.93 | 0.91 | 16 | 2.3 | 0.37 | 35 | 44 | 26 |
| Pillistfer | EL6 | 77 | 28 | 2.9 | 0.917 | 1.00 | 0.92 | 0.92 | 16 | 3.4 | 0.44 | 4.7 | 8.4 | 6.7 |
| llafegh 009 | EL-AN4) | 130 | 38 | 4.5 | 0.906 | 1.00 | 0.87 | 0.80 | 16 | 3.0 | 0.37 | 8.1 | 13 | 12 |

Table 2. Abundances of cosmogenic noble gases and cosmic-ray exposure ages¹⁾.

 $P^3 = F^3 [2.09 - 0.43(^{22}\text{Ne}/^{21}\text{Ne})_c]$

 $P^{21} = 1.61 F^{21} [21.77(^{22} \text{Ne}/^{21})_c - 19.32]^{-1}$

 $P^{38} = F^{38}[0.125 - 0.071(^{22}\text{Ne}/^{21}\text{Ne})_c]$ (P^n in units of 10^{-8} cm³ STP/g per Myr; *n*: nuclide)

¹⁾ Uncertainties for cosmic-ray exposure ages are estimated to be 15% including 10% errors for cosmogenic gas contents and their production rates.

²⁾ Assumed to be entirely cosmogenic except for ALH-77295 (see text).

³⁾ An intermediate composition between EH- and EL-chondrites.

⁴⁾ An impact-melt rock of EL-chondrite.

Cosmogenic ³He, ²¹Ne and ³⁸Ar concentrations listed in Table 2 were calculated by the following assumptions: Obtained ³He is entirely cosmogenic except for ALH-77295, in which solar He is subtracted by assuming that isotopic ratios of trapped (t) and cosmogenic (c) components are $({}^{3}\text{He}/{}^{4}\text{He})_{t} = 3.8 \times 10^{-4}$ and $({}^{3}\text{He}/{}^{4}\text{He})_{c} =$ 0.25, respectively. Radiogenic ⁴He content would be negligible compared to more abundant solar ⁴He in ALH-77295, because bulk ⁴He contents in solar-Ne free E-chondrites are generally lower than 1.5×10^{-6} cm³ STP/g (e.g., Crabb and Anders, 1981; Wacker and Marti, 1983). Except for four samples (ALH-77295, Y-691, -74370 and -791790), measured ²¹Ne/²²Ne ratios are regarded as cosmogenic ones because their 20 Ne/ 22 Ne ratios are within a range of cosmogenic Ne (Fig. 1a). Among ordinary chondrites, a cosmogenic ²⁰Ne/²²Ne ratio is in a narrow range irrespective of ²¹Ne/²²Ne, so the cosmogenic ²¹Ne/²²Ne ratio is generally calculated from a point having the 20 Ne/ 22 Ne ratio of 0.85-0.90 on a mixing line of their trapped and cosmogenic Ne. However, the cosmogenic ²⁰Ne/²²Ne ratios for E-chondrites seem to be correlated with the 21 Ne/ 22 Ne ratios (Fig. 1b). Hence, we tentatively calculated the cosmogenic ²¹Ne/²²Ne ratio from an intersection between a mixing line of trapped and measured Ne, and the least squares fitting line for the E-chondrites having cosmogenic ²⁰Ne/²²Ne. Trapped Ne for Y-691, -74370 and -791790 is assumed to be atmospheric because stepped pyrolysis for some bulk E-chondrites revealed that their isotopic ratios of Ne plot along a mixing line between cosmogenic and air-like Ne (Okazaki R., unpublished data). Huss et al. (1996) also reported that Abee HF/HCl residues contain trapped Ne intermediate between Ne-P1 and Ne-A1, close to atmospheric Ne. For ALH-77295, which includes solar Ne, a least squares fitting to the data points including literature ones (Wieler *et al.*, 1985) is applied to get an intersection. Using these cosmogenic ²¹Ne/²²Ne and (²¹Ne/²²Ne)_t=0.03, trapped ²¹Ne is subtracted from measured ²¹Ne concentrations. For the calculation of cosmogenic ³⁸Ar, (³⁸Ar/³⁶Ar)_t=0.188, and (³⁸Ar/³⁶Ar)_c=1.5 are assumed for all samples.

3.2. Weathering

We performed duplicate measurements for Y-74370 in order to see the degree of sample heterogeneity due to small sample weights. These specimens are prepared from the same block (Y-74370, 67). Isotopic ratios of He, Ne and Ar, and Ne contents are in good agreement between Y-74370 (A) and (B), which suggests that the contribution of sample heterogeneity is not significant. Concentrations of He and Ar, however, varied between the specimens with 2- or 3-fold differences (Table 1).

As mentioned, ³He is almost cosmogenic, while ³⁶Ar and ³⁸Ar are mixtures of cosmogenic and trapped components. Cosmogenic ³⁶Ar and ³⁸Ar in meteorites are produced mainly from Ca and Fe via spallation reactions with galactic cosmic-rays. Kallemeyn and Wasson (1986) reported that weathered E-chondrites tend to be depleted in Ca, Mn, Na, K and Fe. The serious discrepancies in Ar concentrations are reasonably explained by that host phases for cosmogenic Ar are easy to be oxidized and susceptible to weathering. Such a weathering effect has been reported by previous studies (*e.g.*, Gibson and Bogard, 1978; Patzer and Schultz, 1999).

Cosmogenic ³He is produced from O, Mg, Si, Fe and other heavy elements. The ratio of production rate of cosmogenic ³He from metal to that from silicate is estimated $P_{(metal)}^{3}/P_{(silicate)}^{3}=0.66$ from geometric cross-sections (Bogard and Cressy, 1973). For cosmogenic ²¹Ne, the production ratio of metal to silicate is calculated $P_{(metal)}^{21}/P_{(silicate)}^{21}=0.04$ from elemental production rates in Schultz and Freundel (1985) by assuming that a silicate composition is MgSiO₃. The production ratio $P_{(metal)}^{3}/P_{(silicate)}^{3}$ is $10 \times larger than P_{(metal)}^{21}/P_{(silicate)}^{21}$, indicating the fraction of $P_{(metal)}^{3}$ is significant. Therefore, the discrepancy in concentrations of cosmogenic ³He as well as cosmogenic Ar is thought to come from weathering of metal and sulfides.

On the other hand, ⁴He and ⁴⁰Ar are radiogenic. Parent nuclides for radiogenic ⁴He and ⁴⁰Ar are U/Th and K, respectively. Potassium is susceptible to weathering (Gibson and Bogard, 1978; Kallemeyn and Wasson, 1986). The parent elements U and Th are heterogeneously distributed (Bogard *et al.*, 1983). Thus, the discrepancies in concentrations of ⁴He and ⁴⁰Ar between Y-74370 (A) and (B) would be caused by sample heterogeneity and weathering, respectively, or both.

In this paper, we will use the results for the specimen Y-74370 (B), because more abundant He and Ar suggest that there is little weathering effect on the specimen (B). By stepped crushing and heating experiments (Okazaki R., unpublished data) for other samples using larger sample sizes (with 70–100 mg), it is confirmed that cosmogenic ²¹Ne contents in the sample are essentially identical.

3.3. Cosmic-ray exposure age

A cosmic-ray exposure age (T^n) is calculated from the following equation:

$$T^n = A^n / P^n , \qquad (1)$$

where A^n and P^n are concentration and a production rate of cosmogenic stable nuclide *n*, respectively. The production rate depends on shielding conditions as well as energy and composition of incident cosmic-ray particles and a chemical composition of a target meteoroid. Eugster (1988) gave the production rates for chondrites considering these effects by the following equation:

$$P^n = F^n \cdot P^n{}_L, \qquad (2)$$

where P_L^n is a production rate of nuclide *n* in L-chondrites and it is a function of a cosmogenic $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ ratio which is used to correct the shielding condition. F^n is a correction factor for composition of target elements of a meteoroid. To determine precise exposure ages, it is necessary to get appropriate correction factors for the chemical composition and the shielding effect for each sample.

The cosmogenic $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ ratio also depends on chemical compositions of samples. Reedy *et al.* (1979) has reported that the reaction ${}^{24}\text{Mg}(n,\alpha){}^{21}\text{Ne}$ predominantly controls the correlation between $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ and the shielding depth. It is suspected that $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ of E-chondrites would be less sensitive to the shielding depth due to their low Mg contents compared to ordinary chondrites. We estimated the influence of the chemical composition on the correlation between $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ and the shielding depth. We used elemental production rates for a 2π geometry (Hohenberg *et al.*, 1978) in the calculation, and assumed that the difference in the geometry is not significant to discuss the influence of the difference in bulk chemistries on the depth profiles of $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$. Bulk elemental compositions of each L- and E-chondrite are from Wasson and Kallemeyn (1988). The estimation shows no significant difference in the depth profile of $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$



Fig. 2. Cosmogenic ${}^{21}Ne/{}^{22}Ne$ vs. shielding depth. Unit for the shielding depth is in g/cm^2 .

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between L- and EL-chondrites (Fig. 2), while there is a slight difference between Land EH-chondrites. For example, $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ estimated for ALH-77295 and Y-791790 is 0.91 (Table 2), which corresponds to $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c=0.92$ at the same shielding depth within L-chondrite. However, the difference in $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ between L- and EH-chondrites is only 1%, which may be within an uncertainty of $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ estimated by the procedure mentioned earlier. Thus, the difference in $({}^{21}\text{Ne}/{}^{22}\text{Ne})_c$ due to chemical compositions can be ignored, and uncertainties for production rates of ³He, ²¹Ne and ³⁸Ar are assumed to be 10%.

As mentioned above, cosmogenic He and Ar are easily lost by weathering of their host phases. In addition to the weathering affects, diffusive loss of He would cause a short ³He-age. Compared to the cosmogenic ³He, cosmogenic ²¹Ne is less sensitive to diffusive loss, although ²¹Ne-loss from plagioclase has been reported (Bogard and Cressy, 1973). Considering these reasons, exposure ages based on cosmogenic ²¹Ne is more reliable than those on cosmogenic ³He and ³⁸Ar. We use exposure ages based on cosmogenic ²¹Ne in the following discussion.

Cosmogenic gas contents and cosmic-ray exposure ages are listed in Table 2 along with their production rates. We assume single-stage 4π exposures to galactic cosmic rays. Uncertainties for concentrations and production rates for cosmogenic noble gases are estimated to be 10%, so those for cosmic-ray exposure ages are 15%. Sample heterogeneity and gas loss due to weathering are not considered for the uncertainties. In this study, the correction factors F^n were calculated following the procedures by Eugster (1988). In the calculation, measured chemical compositions listed in Table 3 were used, because EH6 has not been included in chemical compositions (Wasson and Kallemeyn, 1988) used in Eugster (1988). The chemical composition of Y-793225 by Lin and Kimura (1998) is different from that determined in this study (Table 3), and shows a extremely low Fe/Mg atomic ratio (0.17) compared to those of other E-chondrites (*e.g.*, EH: 1.19; EL: 0.68; Wasson and Kallemeyn, 1988). So we will adopt our result for Y-793225.

The noble gas data for two EH6 and one EH/EL6 chondrites are the first report on high petrologic types of EH-chondrites and a newly recognized Echondrite having an intermediate composition of (Mg, Mn, Fe)S solid solution between EH- and EL-chondrites (Lin and Kimura, 1998). Petrologic and mineralogical studies by Lin and Kimura (1998) indicate that Y-8414 and -86004 are paired. Our noble gas data shows that isotopic ratios and concentrations of Ne are in good agreement between these two meteorites. The identical exposure ages (10 Myr) supports their potential pairing. Thus, Y-8414 and -86004 should be regarded to be paired. Another possible pair is two E3 meteorites, Y-792959 and -793161. They have identical mineral compositions, and were recovered at a close location in Antarctica (Yanai and Kojima, 1995), which suggests that they might have fallen as a meteorite shower with many fragments. Their isotopic ratios of He, Ne and Ar are identical. Although there are some differences in concentrations, especially in ³He, ⁴He and ⁴⁰Ar contents, such differences in noble gas abundances are also observed in Y-74370 (Table 1). Thus, we will treat Y-792959 and -793161 as a pair in the following discussion, according to their mineralogical and petrologic

| | Y-691 ^{1), 2)} (average) | Y-791790 ³⁾ | Y-79295 9²⁾ | Y-793161 ²⁾ | ALH-77295 ⁴⁾ | Y-74370 ³⁾ | Y-74370 ²⁾ | Y-74370 (average) |
|-----------------|--------------------------------------|------------------------|-------------------------------|------------------------|-------------------------|-----------------------|-----------------------|----------------------|
| Si | 16.97 | 19.85 | 15.09 | 15.50 | 16.69 | 17.03 | 15.96 | 16.49 |
| AI | 1.55 | 1.02 | 1.20 | 0.99 | 0.95 | 1.08 | 1.33 | 1.20 |
| Fe | 28.91 | 20.16 | 28.10 | 27.85 | 30.60 | 25.02 | 27.74 | 26.38 |
| Mn | 0.19 | 0.19 | 0.21 | 0.19 | 0.21 | 0.16 | 0.18 | 0.17 |
| Mg | 11.82 | 13.71 | 10.69 | 10.65 | 11.06 | 12.01 | 10.98 | 11. 4 9 |
| Ca | 0.92 | 0.24 | 0.90 | 0.82 | 0.90 | 0.63 | 0.79 | 0.71 |
| Na | 0.62 | 0.91 | 0.42 | 0.57 | 0.58 | 0.96 | 0.55 | 0.75 |
| ĸ | 0.06 | 0.09 | 0.04 | 0.02 | 0.06 | 0.01 | 0.07 | 0.04 |
| 0** | 30.9 | 37.3 | 36.8 | 36.8 | 31.8 | 38.6 | 35.8 | 37.2 |
| Ni | 1.71 | 1.21 | 1.41 | 1.40 | 1.83 | 0.32 | 1.51 | 0.91 |
| Co | 0.08 | - | 0.059 | 0.047 | 0.09 | 0.003 | 0.051 | 0.027 |
| Ti | 0.05 | 0.04 | 0.04 | 0.05 | 0.07 | 0.04 | 0.07 | 0.05 |
| Cr | 0.31 | 0.30 | 0.23 | 0.29 | 0.27 | 0.16 | 0.29 | 0.23 |
| S | 5.95 | 5.03 | 4.84 | 4.82 | 4.92 | 3.98 | 4.66 | 4.32 |
| Total cation | 69.1 | 62.7 | 63.2 | 63.2 | 68.2 | 61.4 | 64.2 | 62.8 |
| | | | | | | | | |
| F ³ | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 | 1.00 |
| F ²¹ | 0.85 | 0.97 | 0.78 | 0.77 | 0.81 | 0.85 | 0.80 | 0.82 |
| F ³⁸ | 0.87 | 0.65 | 1.06 | 0.88 | 0.98 | 0.66 | 0.91 | 0.7 8 |

Table 3. Bulk chemical composition^{*} and chemical correction factors F^n .

| | Y-8414 ⁵⁾ | Y-8 6004⁵⁾ | Y-793225 ³⁾ | Y-793225 ⁵⁾ | Pillistfer ^{6), 7), 8)} (average) | llafegh 009 ⁹⁾ | Average of EH- chondrite ¹⁰⁾ | Average of EL- chondrite ¹⁰⁾ |
|--------------|----------------------|------------------------------|------------------------|------------------------|---|---------------------------|--|--|
| Si | 24.07 | 24.07 | 20.81 | 25.09 | 18.62 | 17.81 | 16.70 | 18.60 |
| AI | 1.19 | 1.19 | 2.16 | 0.46 | 1.15 | 0.87 | 0.81 | 1.05 |
| Fe | 17.48 | 17.48 | 13.50 | 7.76 | 27.78 | 30.01 | 29.00 | 22.00 |
| Mn | 0.18 | 0.18 | 0.11 | 0.03 | 0.02 | 0.21 | 0.22 | 0.16 |
| Mg | 13.73 | 13.73 | 12.99 | 19.92 | 12.63 | 12.43 | 10.60 | 14.10 |
| Ca | 0.18 | 0.18 | 0.96 | 0.36 | 0.44 | 0.56 | 0.85 | 1.01 |
| Na | 1.07 | 1.07 | 1.42 | 0.29 | 0.59 | 0.49 | 0.68 | 0.58 |
| κ | 0.10 | 0.10 | 0.17 | 0.02 | 0.07 | 0.03 | 0.08 | 0.07 |
| 0** | 36.9 | 36.9 | 41.7 | 42.4 | 33.5 | 32.4 | 28.0 | 31.0 |
| Ni | 1.06 | 1.06 | 0.83 | 0.31 | 1.68 | 1.79 | 1.75 | 1.30 |
| Co | 0.049 | 0.049 | - | 0.015 | 0.080 | 0.082 | 0.084 | 0.067 |
| Ti | 0.02 | 0.02 | 0.02 | 0.07 | - | 0.07 | 0.05 | 0.06 |
| Cr | 0.18 | 0.18 | 0.16 | 0.45 | 0.14 | 0.13 | 0.32 | 0.31 |
| S | 3.77 | 3.77 | 1.54 | 2.76 | 3.29 | 3.15 | 5.80 | 3.30 |
| Total Cation | 63.1 | 63.1 | 96.4 | 99.9 | 66.5 | 67.6 | 94.9 | 93.6 |
| 3 | | | 1.00 | 1.00 | | | 1.00 | 1.00 |
| г 21 | 1.00 | | 1.00 | 1.00 | | 1.00 | 1.00 | 1.00 |
| _₩ | 1.00 | 0.88 | 0.93 | 1.30 | 0.92 | 0.87 | 0.78 | 0.96 |
| F P | 0.60 | 0.69 | 0.91 | 0.33 | 0.92 | 0.80 | 0.98 | 0.89 |

 $F^n = Q^n_{\text{sample}} / Q^n_{\text{L-chondrite}}$ (*n*: nuclide)

 $Q^{3} = 0.0174[Ti + Cr + Mn + Fe + Ni] + 0.0266(100 - [Ti + Cr + Mn + Fe + Ni)]$

 $Q^{21} = 1.63[Mg] + 0.6[A1] + 0.32[Si] + 0.22[S] + 0.07[Ca] + 0.021[Fe + Ni]$

 $Q^{38} = 1.58$ [Ca] + 0.086[Fe + Ni] + 0.33[Ti + Cr + Mn] + 11[K]

([X]: concentration of element X as weight fraction; Q^n in units of 10^{-8} cm³ STP/g per Myr)

* The elemental abundances are in weight percent.

** Oxygen content is calculated by subtracting total cation from 100%.

References: 1) Shima (1974); 2) Yanai and Kojima (1995); 3) this study; 4) Jarosewich (1990); 5) Lin and Kimura (1998); 6) Dyakonova (1968); 7) Jarosewich and Mason (1969); 8) Von Michaelis et al. (1969); 9) McCoy et al. (1995); 10) Wasson and Kallemeyn (1988).

similarities, and their near sampling locations. Cosmic-ray exposure ages of 10 and 44 Myr are adopted for Y-8414 and -86004, and Y-792959 and -793161, respectively.

3.4. Distribution of cosmic-ray exposure ages of E-chondrites

Figure 3 shows distribution of the cosmic-ray exposure ages of twenty nine E-chondrites studied in this work and previously reported ones. Present study adds new data for seven E-chondrites on the distribution proposed by Crabb and Anders (1981). In addition, recently reported four E-chondrites, LEW-87223 (Schultz et al., 1998), Qingzhen (Shima et al., 1973), Parsa (Murty, 1992) and Kishima (Nagao et al., 1993) are also plotted. Kishima meteorite has been reported as EL-chondrite (Nagao et al., 1993), but recently it was classified as EL6 (reported as Kijima meteorite; Okada and Shima, 1998).

Crabb and Anders (1981) reported that exposure ages for E-chondrites have a bimodal pattern and there are systematic differences correlated with petrologic types. Most E4 chondrites show T^{21} below 16 Myr, while T^{21} for E6 chondrites are above 30 Myr. They proposed that there was a single E-chondrite parent body,



Fig. 3. Distribution of cosmic-ray exposure ages based on cosmogenic ²¹Ne. Average exposure ages of our results and literature ones (Crabb and Anders, 1981) are 7.6 and 3.8 Myr for Pillistfer and Y-691, respectively. AN means exceptional stones in their petrologic types: (1) Abee and Adhi Kot have been reported as EH4 chondrites (Sears et al., 1982), and later as EH impact-melt breccias (Rubin, 1995; Rubin and Scott, 1997). (2) LEW-87223 is interpreted as an anomalous E3 chondrite (Zhang et al., 1995) or an anomalous EL-chondrite (Zhang and Sears, 1996). (3) McCoy et al. (1995) has reported that Ilafegh 009 and Happy Canyon are an impact-melt rock and an impact-melt breccia of EL-chondrites, respectively.

which contained predominantly a single petrologic type on the scale of individual impacts, based on the systematic distribution of cosmic-ray exposure ages. Contrary to the report of Crabb and Anders (1981), no systematic difference in the exposure ages with respect to petrologic types is observed (Fig. 3). Exposure ages obtained for type 3 and 4 E-chondrites cluster around 3.5 Myr and 45 Myr, while those for higher petrologic types of E-chondrites do around 10 Myr except for Y-793225 (EH/EL6) with 45 Myr (Table 2). The peaks around 5 and 40 Myr contain both E3 and E6 chondrites. Patzer and Schultz (1998) also reported no relation between exposure ages and petrologic types. In addition, the present study discovered that both EH- and EL-chondrites are found in both peaks, that is, there is little difference in the exposure age distribution between EH- and EL-subclasses. Thus, the exposure age distribution hardly give information on the structure of the parent body (or bodies) of E-chondrites. The most simple explanation for the exposure age distribution is that the parent body (or bodies) had been collapsed prior to collisions which threw out meter-sized E-chondritic materials.

Exposure ages reflect the last irradiation episode in space after fragmentation into meter-sized objects. Therefore, the exposure age of E-chondrites does not imply their origin and the structure of their parent body, unlike some arguments previously presented (*e.g.*, Crabb and Anders, 1981; Kong *et al.*, 1997).

4. Summary

Light noble gas data for eleven E-chondrites are presented. Based on the noble gas signatures and their exposure ages (T^{21}) , potential pairing between Y-8414 (EH6) and -86004 (EH6) is suggested. Cosmic-ray exposure ages for EH- and EL-chondrites have correlation with respect to neither their petrologic types nor chemical classification. The exposure age distribution would not reflect the original structure of E-chondrite parent body.

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References

- Akridge, D. G. and Sears, D. W. G. (1999): The gravitational and aerodynamic sorting of meteoritic chondrules and metal: Experimental results with implications for chondritic meteorites. J. Geophys. Res., 104 (E5), 11853-11864.
- Baedecker, P. A. and Wasson, J. T. (1975): Elemental fractionations among enstatite chondrites. Geochim. Cosmochim. Acta, **39**, 735-765.

- Bishoff, A., Palme, H., Geiger, T. and Spettel, B. (1992): Mineralogy and Chemistry of the EL-chondritic melt rock Ilafegh-009. Lunar and Planetary Science XXIII. Houston, Lunar Planet Inst., 105-106.
- Biswas, S., Walsh, T., Bart, G. and Lipschutz, M. E. (1980): Thermal metamorphism of primitive meteorites-XI. The enstatite meteorites: origin and evolution of a parent body. Geochim. Cosmochim. Acta, 44, 2097-2110.
- Bogard, D. D. and Cressy, P. J. (1973): Spallation production of ³He, ²¹Ne and ³⁸Ar from target elements in the Bruderheim chondrite. Geochim. Cosmochim. Acta, **37**, 527–546.
- Bogard, D. D., Unruh, D. M. and Tatsumoto, M. (1983): ⁴⁰Ar/³⁹Ar and U-Th-Pb dating of separated clasts from the Abee E4 chondrite. Earth Planet. Sci. Lett., **62**, 132–146.
- Clayton, R. N., Mayeda, T. K. and Rubin, A. E. (1984): Oxygen isotopic compositions of enstatite chondrites and aubrites. Proc. Lunar Planet. Sci. Conf., 15th, Pt. 1, C245-C249 (J. Geophys. Res., 89 Suppl.).
- Crabb, J. and Anders, E. (1981): Noble gases in E-chondrites. Geochim. Cosmochim. Acta, 45, 2443-2464.
- Crabb, J. and Anders, E. (1982): On the siting of noble gases in E-chondrites. Geochim. Cosmochim. Acta, 46, 2351-2361.
- Dyakonova, M. I. (1968): Khimicheskii sostav semi khondritov raznykh tipov. Meteoritika, 28, 131-137.
- El Goresy, A., Yabuki, H., Ehlers, K., Woolum, D. and Pernicka, E. (1988): Qingzhen and Yamato-691: A tentative alphabet for the EH chondrites. Proc. NIPR Symp. Antarct. Meteorites, 1, 65-101.
- Eugster, O. (1988): Cosmic-ray production rates for ³He, ²¹Ne, ³⁸Ar, ⁸³Kr, and ¹²⁶Xe in chondrites based on ⁸¹Kr-Kr exposure ages. Geochim. Cosmochim. Acta, **52**, 1649–1662.
- Eugster, O., Michel, Th., Niedermann, S., Wang, D. and Yi, W. (1993): The record of cosmogenic, radiogenic, fissiogenic and trapped noble gases in recently recovered Chinese and other chondrites. Geochim. Cosmochim. Acta, 57, 1115–1142.
- Gibson, E. K., Jr. and Bogard, D. D. (1978): Chemical alterations of the Holbrook chondrite resulting from terrestrial weathering. Meteoritics, 13, 277-297.
- Hohenberg, C. M., Marti, K., Podosek, F. A., Reedy, R. C. and Shirck, J. R. (1978): Comparisons between observed and predicted cosmogenic noble gases in lunar samples. Proc. Lunar Planet. Sci. Conf., 9th, 2311-2344.
- Huss, G. R., Lewis, R. S. and Hemkin, S. (1996): The "normal planetary" noble gas component in primitive chondrites: Compositions, carrier, and metamorphic history. Geochim. Cosmochim. Acta, 60, 3311-3340.
- Jarosewich, E. (1990): Chemical analyses of meteorites: A compilation of stony and iron meteorite analyses. Meteoritics, 25, 323-337.
- Jarosewich, E. and Mason, B. (1969): Chemical analyses with notes on one mesosiderite and seven chondrites. Geochim. Cosmochim. Acta, 33, 411-416.
- Kaczaral, P. W., Dennison, J. E., Verkouteren, R. M. and Lipschutz, M. E. (1988): On volatile/mobile trace element trends in E3 chondrites. Proc. NIPR Symp. Antarct. Meteorites, 1, 113-121.
- Kallemeyn, G. W. and Wasson, J. T. (1986): Compositions of enstatite (EH3, EH4, 5 and EL6) chondrites: Implications regarding their formation. Geochim. Cosmochim. Acta, 50, 2153– 2164.
- Keil, K. (1968): Mineralogical and chemical relationships among enstatite chondrites. J. Geophys. Res., 73, 6945–6976.
- Keil, K. (1989): Enstatite meteorites and their parent bodies. Meteoritics, 24, 195-208.
- Kong, P., Mori, T. and Ebihara, M. (1997): Compositional continuity of enstatite chondrites and implications for heterogeneous accretion of the enstatite chondrite parent body. Geochim. Cosmochim. Acta, 61, 4895-4914.
- Lin, Y. and Kimura, M. (1998): Petrographic and mineralogical study of new EH melt rocks and a new enstatite chondrite grouplet. Meteorit. Planet. Sci., 33, 501-511.

Marti, K. (1967): Trapped Xe and the classification of chondrites. Earth Planet. Sci. Lett., 2, 193-196.

- McCoy, T. J., Keil, K., Bogard, D., Casanova, I. and Lindstrom, M. M. (1992): Ilafegh 009: A new sample of the diverse suite of enstatite impact melt rocks. Lunar and Planetary Science XXIII. Houston, Lunar Planet Inst., 869–870.
- McCoy, T. J., Keil, K., Bogard, D. D., Garrison, D. H., Casanova, I., Lindstrom, M. M., Brearley, A. J., Kehm, K., Nichols, R. H., Jr. and Hohenberg, C. M. (1995): Origin and history of impact-melt rocks of enstatite chondrite parentage. Geochim. Cosmochim. Acta, 59, 161-175.
- Murty, S. V. S. (1992): Solar gases in Parsa enstatite chondrite. Lunar and Planetary Science XXIII. Houston, Lunar Planet Inst., 951–952.
- Nagao, K., Miura, Y. N. and Shima, M. (1993): Noble gases in Japanese chondrites: Their gas retention and cosmic-ray exposure histories. J. Mass Spectrom. Soc. Jpn., 41, 191-209.
- Nagao, K., Okazaki, R., Sawada, S. and Nakamura, N. (1999): Noble gases and K-Ar ages of five Rumuruti chondrites Yamato (Y) -75302, Y-791827, Y-793575, Y-82002, and Asuka-881988. Antarct. Meteorite Res., 12, 81-93.
- Nehru, C. E., Prinz, M., Weisberg, M. K. and Delaney, J. S. (1984): Parsa: An unequilibrated enstatite chondrite (UEC) with an aubrite-like impact melt clast. Lunar and Planetary Science XV. Houston, Lunar Planet. Inst., 597-598.
- Okada, A. and Shima, M. (1998): Kijima meteorite: A new fiding of free diopside-bearing EL6 chondrite. 17th general meeting of the International Mineralogical Association, A44.
- Patzer, A. and Schultz, L. (1998): The exposure age distribution of enstatite chondrites. Meteorit. Planet. Sci., 33, A120-A121.
- Patzer, A. and Schultz, L. (1999): Do weathering effects influence cosmic ray exposure ages of enstatite chondrites? Workshop on "Extraterrestrial Materials from Cold and Hot Desserts". (Held at Kwa-Maritane, Pilanesberg, South Africa, July 6-8, 1999)
- Prinz, M., Nehru, C. E., Weisberg, M. K. and Delaney, J. S. (1984): Type 3 enstatite chondrites: A newly recognized group of unequilibrated enstatite chondrites (UEC's). Lunar and Planetary Science XV. Houston, Lunar Planet. Inst., 653–654.
- Reedy, R. C., Herzog, G. F. and Jessberger, E. K. (1979): The reaction $Mg(n, \alpha)Ne$ at 14.1 and 14.7 MeV: Cross sections and implications for meteorites. Earth Planet. Sci. Lett., 44, 341–348.
- Reynolds, J. H. (1960): Isotopic composition of xenon from enstatite chondrites. Z. Naturforsch., 15a, 1112-1114.
- Rubin, A. E. (1995): Petrologic evidence for collisional heating of chondritic asteroids. Icarus, 113, 156-167.
- Rubin, A. E. and Scott, E. R. D. (1997): Abee and related EH chondrite impact-melt breccias. Geochim. Cosmochim. Acta, 61, 425-435.
- Rubin, A. E., Scott, E. R. D. and Keil, K. (1997): Shock metamorphism of enstatite chondrites. Geochim. Cosmochim. Acta, 61, 847-858.
- Schultz, L., Franke, L. and Kruse, H. (1998): Helium, neon and argon in meteorites a data compilation update 1998.
- Schultz, L. and Freundel, M. (1985): On the production rate of ²¹Ne in ordinary chondrites. Isotopic Ratios in the Solar System, ed. by Centre National D'Etudes Spatiales, Cepadues-Editions, Toulouse, 27-33.
- Sears, D. W. G., Kallemeyn, G. W. and Wasson, J. T. (1982): The compositional classification of chondrites: II The enstatite chondrite groups. Geochim. Cosmochim. Acta, 46, 597–608.
- Sheng, Z., Sallee, W. and Sears, D. W. G. (1982): Trace element data on enstatite chondrite components and the Qingzhen enstatite chondrite. Lunar and Planetary Science XIII. Houston, Lunar Planet. Inst., 718-719.
- Shima, M. (1974): The chemical compositions of the stone meteorites Yamato (a), (b), (c) and (d), and Numakai. Meteoritics, 9, 123-135.
- Shima, M., Shima, M. and Hintenberger, H. (1973): Chemical composition and rare gas content of four new detected Antarctic meteorites. Earth Planet. Sci. Lett., **19**, 246-249.
- Signer, P., Baur, H., Etique, Ph. and Wieler, R. (1983): Light noble gases in 15 meteorites. Meteoritics,

18, 399.

- Van Schumus, W. R. and Wood, J. A. (1967): A chemical-petrological classification for the chondritic meteorites. Geochim. Cosmochim. Acta, 31, 747-765.
- Von Michaelis, H., Ahrens, L. H. and Willis, J. P. (1969): The composition of stony meteorites II. The analytical data and an assessment of their quality. Earth Planet. Sci. Lett., 5, 387–394.
- Wacker, J. F. and Marti, K. (1983): Noble gas components in clasts and separates of the Abee meteorite. Earth Planet. Sci. Lett., 62, 147–158.
- Wang, D. and Xie, X. (1981): Preliminary investigation of mineralogy, petrology and chemical composition of Qingzhen enstatite chondrite. Geochemistry, 1, 69-81.
- Wasson, J. T. and Kallemeyn, G. W. (1988): Compositions of chondrites. Philos. Trans. R. Soc. London, A325, 535-544.
- Wasson, J. T. and Wai, C. M. (1970): Composition of the metal, schreibersite and perryite of enstatite achondrites and the origin of enstatite chondrites and achondrites. Geochim. Cosmochim. Acta, 34, 169-184.
- Wieler, R., Baur, H., Graf, Th. and Signer, P. (1985): He, Ne and Ar in Antarctic meteorites: Solar noble gases in an enstatite chondrite. Lunar and Planetary Science XVI. Houston, Lunar Planet. Inst., 902–903.
- Yanai, K. and Kojima, H. (1995): Catalog of the Antarctic Meteorites. Tokyo, Natl Inst. Polar Res., 230 p.
- Zähringer, J. (1962): Isotopie-Effekt und Häufigkeiten der Edelgase in Steinmeteoriten und auf der Erde. Z. Naturforsch., 17a, 460-471.
- Zähringer, J. (1966): Die Chronologie der Chondriten aufgrund von Edelgasisotopen-Analysen. Meteoritika, 27, 25-40.
- Zähringer, J. (1968): Rare gases in stony meteorites. Geochim. Cosmochim. Acta, 32, 209-237.
- Zhang, Y. and Sears, D. W. G. (1996): The thermometry of enstatite chondrites: A brief review and update. Meteorit. Planet. Sci., 31, 647-655.
- Zhang, Y., Benoit, P. H. and Sears, D. W. G. (1995): The classification and complex thermal history of the enstatite chondrites. J. Geophys. Res., 100, 9417-9438.

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Appendix

Appendix. Samples discussed in this study, and their references for classification and noble gas data.

| Sample | Classification | References of chemical class | References of petrologic type | References of noble gases |
|--------------------|---|------------------------------|-------------------------------|---------------------------|
| Abee | EH imapct-melt breccia | 1 | 2 | 3 |
| Adhi Kot | EH imapct-melt breccia | 1 | 2 | 3 |
| ALH-77295 | EH4 | 4 | 4 | 5 |
| Atlanta | EL6 | 1 | 6 | 3 |
| Bethune | EH4 | 1 | 7 | 3 |
| Blithfield | EL6 | 1 | 6 | 3 |
| Daniel's Kuil | EL6 | 1 | 6 | 3 |
| Happy Canyon | EL impact-melt breccia | 8 | 8 | 3 |
| Hvittis | EL6 | 1 | 6 | 3 |
| Ilafegh 009 | EL impact-melt rock | 9, 10 | 11 | 5 |
| Indarch | EH4 | 1 | 6 | 3 |
| Jajh Deh Kot Lalu | EL6 | 1 | 6 | 3 |
| Khairpur | EL6 | 1 | 6 | 3 |
| Kishima | EL | 12 | 13 | 12 |
| LEW-87223 | anomalous EL3 | 14 | 15 | 16 |
| North West Forrest | EL6 | 1 | 6 | 3 |
| Parsa | EH3 | 1 | 11, 17 | 18 |
| Pillistfer | EL6 | 1 | 6 | 3, 5 |
| Qingzhen | EH3 | 19 | 20 | 21 |
| Saint-Sauveur | EH5 | 1 | 6 | 3 |
| South Oman | EH4/5 | 1 | 6 | 3 |
| St. Mark's | EH5 | 1 | 6 | 3 |
| Y-691 | EH3 | 1 | 22 | 3, 5, 23 |
| Y-74370 | EH4 | 4 | 4 | 5 |
| Y-791790 | EH3 | 7 | 4 | 5 |
| Y-792959 | EH3 | 7 | 4 | 5 |
| Y-793161 | E3 | | 4 | 5 |
| Y-793225 | type 6 chondrite with an intermediate between EH and EL | 24 | 24 | 5 |
| Y-8414 | EH6 | 24 | 24 | 5 |
| Y-86004 | EH6 | 24 | 24 | 5 |
| Yilmia | EL6 | 1 | 6 | 3 |

References: 1) Sears et al. (1982); 2) Rubin and Scott (1997); 3) Crabb and Anders (1981); 4) Yanai and Kojima (1995); 5) this study; 6) Van Schumus and Wood (1967); 7) Rubin et al. (1997); 8) McCoy et al. (1995); 9) McCoy et al. (1992); 10) Bishoff et al. (1992); 11) Prinz et al. (1984); 12) Nagao et al. (1993); 13) Okada and Shima (1998) ; 14) Zhang et al. (1995); 15) Zhang and Sears (1996); 16) Schultz et al. (1998); 17) Nehru et al. (1984); 18) Murty (1992); 19) Sheng et al. (1982); 20) Wang and Xie (1981); 21) Eugster et al. (1993); 22) El Goresy et al. (1988); 23) Shima et al. (1973); 24) Lin and Kimura (1998).