Geochemistry and preliminary Sr-Nd isotopic data on the Neoproterozoic granitoids from the Bantoum area, west Cameroon: evidence for a derivation from a Paleoproterozoic to Archean crust

Charles Nzolang¹, Hiroo Kagami¹, Jean Paul Nzenti² and François Holtz³

¹ Graduate School of Science and Technology, Niigata University, Ikarashi 2-nocho, Niigata 950-2181 (nzolang@gs.niigata-u.ac.jp)

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Abstract: The Bantoum area in west Cameroon is composed of migmatitic gneisses associated with parallel strips of amphibolites, quartz-monzonites, biotite-granites, two-mica leucogranites and granitic dikes. Quartz-monzonites are metaluminous (A/ CNK = 0.8-0.9) I-type, biotite-granites are peraluminous (A/CNK = 1.0-1.10) I-type, leucogranites are peraluminous (A/CNK=1.14) S-type granitoids. All are hyperpotassic rocks defining a calc-alkaline trend. Quartz-monzonites gave an Rb-Sr isochron age of 720±61 Ma assumed to be a mixing age. The thermometry estimated from major elements and zircon saturation indicate that the biotite-granites crystallized from high temperature melts (812-866°C) whereas leucogranites crystallized from low temperature melts (719-745°C). The trace element distribution diagrams are characterized by an enrichment in LILE and LREE (5 < La_N/Sm_N < 17), with negative Nb, Ta, Sr and Ti anomalies. Model initial 87Sr/86Sr ratios (620 Ma) are 0.707614-0.708363 for quartz-monzonites, 0.711242-0.713784 for biotite-granites, and 0.715835 for leucogranites. They have highly negative ε_{Nd} (620 Ma) (-19~-11) and T_{DM} model ages ranging from 1.9 to 2.9 Ga. These geochemical and isotopic features imply that the granites are generated at different temperatures and from different crustal materials; they are the witnesses of the recycling of a Paleoproterozoic to Archean crust with minor inputs of juvenile magmas during the Pan-African orogeny. Chemical similarities between gneisses and some biotite-granites suggest that the partial melting of these gneisses may have contributed to the formation of granites.

key words: granite, geochemistry, Sr-Nd isotope, crustal recycling, west-Cameroon

1. Introduction

Since geochronological (Rb-Sr) works from Lasserre (1966, 1967), the Precambrian basement rocks in Cameroon was separated into the Archean Ntem complex (northwestern edge of Congo craton) in the south and the remaining territory belonging to the Pan-African "mobile zone"; the latter is now referred to as Central African fold belt (CAFB) (Penaye *et al.*, 1993; Toteu *et al.*, 2001) or Pan-African north-equatorial fold belt (Nzenti, 1987). The Ntem complex is essentially composed of charnockitic gneisses and various granitoids. Ages reported from recent works on that complex

² Département des Sciences de la Terre, Université de Yaoundé I, B.P. 812, Yaoundé, Cameroon ³ Institut für Mineralogie, Universität Hannover, Callinstraβe 3, D-30167 Hannover, Germany

range from 2.3 Ga for syenites (Pb-Pb zircon; Tchameni et al., 2001), 2.6 Ga for K-rich granitoids (Pb-Pb zircon; Tchameni et al., 2000) to 2.9 Ga for charnockites (U-Pb zircon; Toteu et al., 1994b). The CAFB in its Cameroon part (Fig. 1) is dominated by 500–700 Ma formations. It used to be roughly divided into 3 main lithotectonic units namely; (a) Paleoproterozoic basement predominantly composed of heterogeneous migmatitic gneisses; the study area is a part of this unit, (b) Meso- to Neoproterozoic volcano-sedimentary basins that are deformed and metamorphosed into schists and high-grade gneisses, and (c) Pan-African granitoids whose emplacement ages range from the early stage of the deformation (orthogneisses) to the late uplift stages of the belt (post-tectonic subcircular massifs), and that cross-cut the latter two units (Nzenti et al., 1988; Penaye et al., 1989; Toteu et al., 2001). However, using only petrological criteria, it is still difficult to distinguish between Neoproterozoic and Paleoproterozoic ages among high-grade rocks as remnants of Paleoproterozoic formations are common all over the fold belt.

Within the Precambrian basement in Cameroon, granitic rocks constitute the major component of the Pan-African terrains (Fig. 1) but they are still poorly known. Based on the compilation works from Bessoles and Trompette (1980), these granitic rocks are classified into two types, namely: the "circumscribed granites" that form wide massifs with well-delimited boundaries distinguishing them from their country rocks; they are mainly intrusive and show a homogeneity in texture and composition, and "not circumscribed granites" usually concordant with country rocks, the transition between the latter and granites being difficult to decipher; they show great variation in texture and structure and frequently contain enclaves. Bessoles and Trompette (1980) assumed that circumscribed granites are probably the more evolved terms of not circumscribed granites.

Some recent investigations have been carried out on some major magmatic complexes in west Cameroon: Ngondo granitic complex (Tagné Kamga, 1994; Tagné Kamga et al., 1999), Bangangté syenites (Tchouankoué, 1992), Batié granites (Talla, 1995), Fomopéa granites (Kwékam, 1993), Bandja charnockitic massif (Nguiessi Tchankam et al., 1997). All are interpreted by the above authors to be metaluminous, calc-alkaline and hyperpotassic syntectonic plutons, formed from mantle-derived magmas with little crustal contamination. Their emplacement ages (Rb-Sr whole rock) vary from 575 to 657 Ma; an U-Pb (zircon) age of 640 ± 15 Ma is obtained on the Bandja charnockites (Nguiessi Tchankam and Vialette, 1994). But little is known concerning the so-called heterogeneous gneisses that are really associated with important exposures of granitoids as we will describe below in the Bantoum area in west Cameroon.

The sole work including this study area is reconnaissance survey of Weecksteen (1957) in which formations of the study area are simply mapped as hornblende-biotite gneisses. We believe that more detailed studies should be carried out on these heterogeneous migmatitic unit for a better understanding of their petrogenesis and the possibly link between gneisses and associated granitoids. In this paper we present the geochemistry and preliminary Rb-Sr and Sm-Nd isotopic data on granitoids and migmatitic gneisses from the Bantoum area, and discuss the nature of source materials of these rocks.

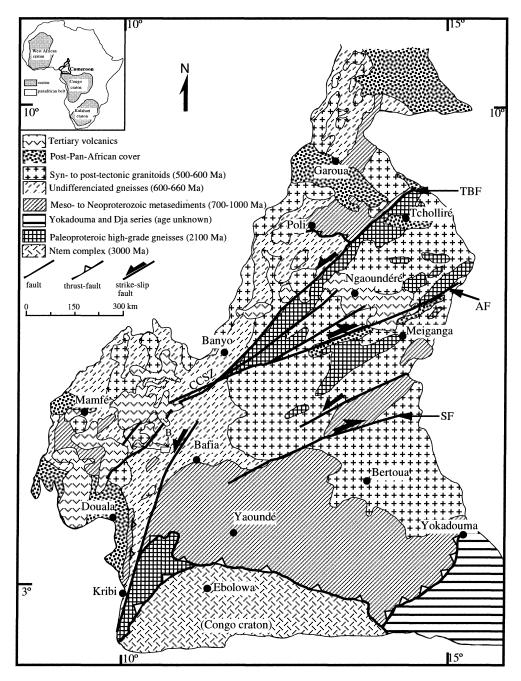


Fig. 1. Geologic sketch map of Cameroon, showing major lithotectonic units (after Soba, 1989; modified by Ngnotué et al., 2000). Ages are from Toteu et al. (2001). AF, Adamaoua fault; CCSZ, central Cameroon shear zone; SF, Sanaga fault; TBF, Tcholliré-Banyo fault. B, study area (Fig. 2).

2. Geological setting

Summarily, two major phases of deformation (D_1 and D_2) are recognized in the CAFB (Ngako *et al.*, 1991; Nzenti *et al.*, 1992), each of them accompanied by important magmatic activity. This magmatism is dominated by Neoproterozoic plutonism. In northern Cameroon, pre- to syn- D_1 intrusions dated at 630 Ma (U-Pb zircon; Toteu *et al.*, 1987) are mainly calc-alkaline, basic to intermediate (diorites, granodiorites, tonalites) and they evoke an accretional domain. In the eastern part, Soba (1989) reported widespread diorites and peraluminous granites dated at 621–614 Ma. Syn- to late D_2 granitoids are represented by calc-alkaline to sub-alkaline anatectic granites, and syenitic and charnockitic plutons.

The D₂ structural evolution is dominated by ENE shears, among which the Adamaoua fault or Central Cameroon Shear Zone (CCSZ, Ngako *et al.*, 1991) is the most important. The CCSZ is a bulk N70°E dextral wrench fault separating the CAFB in Cameroon into two main structural domains: a northern domain dominated by N-S to NW-SE trends, and a southern domain dominated by E-W structures. This lineament plays a key-role in controlling the main geometrical features of the structures of the fold belt and is assumed by Ngako *et al.* (1991, 1992) to be an ultimate consequence of an important E-W crustal shortening during the Pan-African orogenesis. Between these two structural domains the transition corresponds to a N30°E mega shear zone that is marked out in west Cameroon with, besides the numerous syntectonic Pan-African plutons, also the alignment of volcanoes and anorogenic plutons that form a part of the Cameroon volcano-tectonic line whose age range from Eocene to the present (Déruelle *et al.*, 1991).

The structural evolution in the Bantoum area comprises two main phases of deformation: the phase D_1 is tangential and dominated by the schistosity and foliation planes S_1 with a mean direction $N30^{\circ}E$, associated with a stretching lineation L_1 and isoclinal folds P_1 ; the phase D_2 which is observed in metamorphic formations essentially is shearing and characterized by shearing planes C_2 often filled with migmatitic leucosomes, and folds P_2 associated with an axial plane schistosity S_2 . As the main structural trends in D_1 and D_2 are close $(N0^{\circ}-N30^{\circ}E)$, these two deformation phases constitute probably two successive stages belonging to a single major deformation phase that has shaped the $N30^{\circ}E$ shear zone in west Cameroon. Granitoids are syn- to late-D1.

The geological map of the Bantoum area is shown in Fig. 2. The northern part is covered with tertiary basaltic trapps. The basement is essentially made up of migmatitic gneisses to which are closely associated: (1) amphibolites visible only in the form of small metric bodies or thin strips interlayered in these gneisses; (2) biotite-granites, two-mica leucogranites and quartz-monzonites occurring as concordant strips of variable widths (5 to 200 m) roughly elongated following the N70°E direction and alternating with strips of surrounding migmatitic gneisses (Fig. 3); (3) small granitic dikes crosscutting the above formations. Elongated rounded enclaves of gneisses are usual in biotite-granites (Fig. 3b), and enclaves of a mafic granulites are locally found in gneisses at the west of the study area. These formations occur mainly on mountain tops or along river beds either as isolated boulders or as large flagstones, and sometimes in small inselberg reliefs through the lateritic cover. The contacts between migmatitic

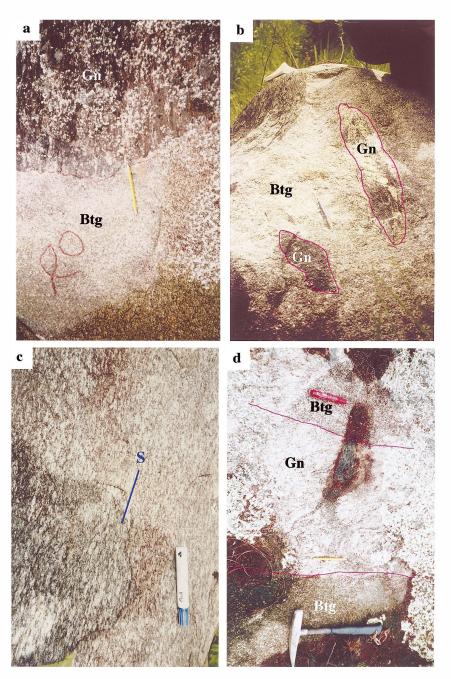


Fig. 3. Field occurrences of granitoids and gneisses in the Bantoum area. a) Contact between biotite-granite and gneiss, with same continuous foliation in both formations (the handwriting was for outcrop identification and has no meaning here). b) Biotite-granite containing elongated enclaves of migmatitic gneiss that are parallel to the foliation. c) Block of quartz-monzonite showing a strong mylonitic foliation. d) Parallel strips of biotite-granites and migmatitic gneiss. Btg, biotite-granite; Gn, gneiss; S, mylonitic foliation.

gneisses and granites are concordant and marked either by the parallelism between the migmatitic foliation and the mylonitic schistosity in the granites, or by the continuation or extension of these structures from one petrographic type to the other one (Fig. 3a). All the granitoids have undergone the same deformations as the gneisses and both share the same structural patterns; thus these granitoids deserve the appellation of "orthogneisses" as defined by Berthé $et\ al.$ (1979) but hereafter we are still using the term granite for the sake of simplification of terminology. Since these granitoids grade into the migmatitic gneisses at the margins and are deformed along with these gneisses, and the presence of gneissic enclaves in granites we assume they could be considered as syn- to late-D₁ intrusives.

3. Petrography

Gneisses are mesocratic and medium-grained and display alternating millimeter to centimeter thick quartzo-feldspathic layers and ferro-magnesian minerals rich layers; this layering is enhanced by migmatization (metatexites) where quartzo-feldspathic leucosomes are well-developed. Their textures are predominantly granoblastic iso- or heterogranular. Amongst these gneisses, amphibole-biotite gneiss is volumetrically the most representative facies and it shows the following mineral association: quartz+ K-feldspar+plagioclase(An20-22)+biotite+hornblende. At places one observes occurrence of small pink garnet crystals, that sometimes cluster to form rounded centimetric grains. Mineral association in these garnet-bearing gneisses is quartz+K-feldspar+plagioclase(An17-24)+biotite+garnet. The occurrence of plagioclase-rich gneiss is more restricted. Its mineral association is plagioclase(An31-35)+quartz+K-feldspar+biotite+garnet; here garnet occurs as small and isolated crystals. It is in this plagioclase-rich gneiss that enclaves of mafic granulites are found. Accessory minerals in all facies are zircon, allanite, apatite, titanite and opaques.

Amphibolites are granoblastic heterogranular with grain size $<4\,\mathrm{mm}$. The mineral assemblage is hornblende + biotite + plagioclase(An29-35) + quartz + K-feldspar, with sphene, zircon, apatite and oxides as accessories. Hornblende forms subhedral to anhedral crystals containing inclusions of euhedral biotite, quartz and accessory minerals. Some biotite crystals include apatite, zircon and sphene. Plagioclase occurs as small subhedral crystals associated with anhedral quartz, both interstitial amidst hornblende and biotite. Sphene crystallizes in general as xenomorph rims around opaque minerals.

Mafic granulites enclaves shows dark-green melanocratic portions with granoblastic texture, essentially composed of hornblende associated with plagioclase (An91-95) and minor amount of biotite and quartz, that grade to mesocratic portions dominated by symplectitic textures comprising the association (\pm hornblende)+plagioclase(An98-100)+clinopyroxene+garnet+quartz and spinel-corundum corona. Accessory phases are ilmenite, magnetite, apatite and zircon.

Biotite-granites (Fig. 4a) are medium-grained and heterogranular. Due to shearing and stretching undergone by these granites, most of their main constituent minerals have been converted into fine aggregates. Porphyroclasts of more resistant minerals (usually feldspars) occur in fine-grained quartz-feldspathic matrix showing the effects of

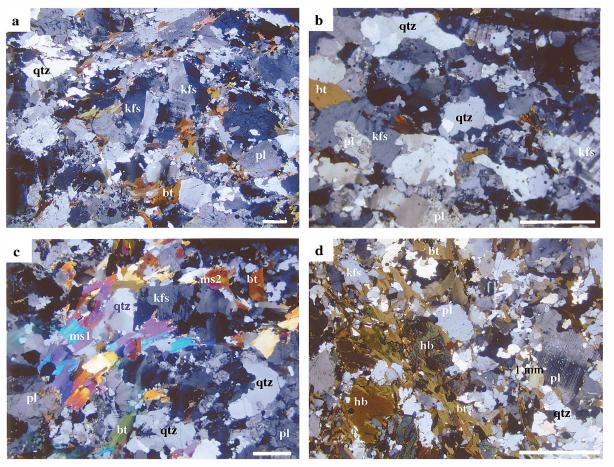


Fig. 4. Photomicrographs (crossed polars) from a) biotite-granite (sample S2.3), b) granitic dike (sample S1.4), c) leucogranite (sample S1.9b); ms1 primary, ms2 secondary muscovite, and d) quartz-monzonite (sample S4.6); all scale bars represent 1 mm. Mineral abbreviations after Kretz (1983).

dynamic recrystallization and strong preferred orientation. Myrmekitic texture is common at the borders of these large resistant crystals of feldspar in contact with recrystallized groundmass. At places primary magmatic texture is still conserved.

Biotite-granites are composed of quartz, K-feldspar, plagioclase (An23-29), biotite; accessory minerals include sphene, allanite, zircon, apatite and opaques. K-feldspar is present as microcline and orthoclase in order of abundance; crystals are dominantly subhedral, rounded or xenomorph; they are perthitic and include small biotite flakes, euhedral plagioclase and quartz. Plagioclase occurs as subhedral or rounded crystals in general elongated following the schistosity; they show sometimes undulose extinction or can be kinked. Some euhedral inclusion-free K-feldspar and plagioclase are randomly oriented, cross-cutting the schistosity. Biotite occurs as flakes of various dimensions that could be clustered, most of which have corroded margins, or as euhedral inclusions in feldspars.

Granitic dikes are fine-grained (Fig. 4b) and composed of anhedral quartz, K-feldspar, plagioclase, biotite, and muscovite. Accessory minerals (opaques, apatite and zircon) are scarce. Some ovoid plagioclase crystals are rimmed by K-feldspar. Biotite flakes in general show corroded margins and are associated with quartz-feldspathic assemblages. Muscovite is xenomorph and crystallizes at the expense of biotite and feldspar.

Two-mica leucogranite is medium-grained, exhibiting granoblastic texture with preferred orientation of feldspar, micas and quartz ribbons. K-feldspar, plagioclase (An14-16), quartz, muscovite and biotite are the main constituents, and zircon, apatite and opaques are accessory phases. K-feldspar, dominantly microcline and perthite is the most abundant mineral and occurs as subhedral to anhedral crystals; some porphyrocrysts contain inclusions of quartz, micas and plagioclase. Quartz forms either as small inclusions in other phases, or large crystals with ondulose extinction, commonly stretched or forming polycrystalline ribbons. Myrmekites are observed as small spots disseminated in some plagioclase or invading K-feldspar and plagioclase from their margins. Using the criteria of shape and relation with other neighboring phases (Lameyre, 1973; Villa et al., 1997), we have distinguished two generations of muscovite (Fig. 4c). Primary muscovite is the most abundant. It occurs as small euhedral inclusions or as well-shaped flakes with sharp margins and grain size ranging from 0.5 to 3.5 mm; the latter are disseminated or clustered and they include only zircon. Secondary muscovite is generally xenomorph with irregular margins growing into feldspar and biotite; they may contain inclusions of biotite, feldspar and quartz. Biotite is less abundant than muscovite.

Quartz-monzonites (Fig. 4d) are leucocratic and fine- to medium-grained showing granoblastic texture. Plagioclase (An30-38) forms either anhedral grains intergrowing with other phases, or big subhedral elongated crystals usually surrounded with small groundmass crystals. K-feldspar is in general rounded, with inclusions of plagioclase, biotite, quartz, apatite and zircon; some ones are perthitic. Biotite flakes are usually aggregated around feldspars or are associated with hornblende to form mafic clusters. Quartz is interstitial or occurs as polycrystalline recrystallized grains. Sphene could be euhedral or xenomorph, forming rims around and also intergrowth with opaques; other accessories are zircon, acicular apatite, allanite and secondary epidote growing at the

expense of hornblende.

4. Geochemistry

4.1. Analytical procedures

Samples weighing 0.5– $1.5\,kg$ were reduced and finely chipped. Afterwards, they were cleaned four times using an ultrasonic bath with ultrapure water. Sample chips were crushed by hand in a tungsten carbide mill, split by quartering and finely ground. For some samples, analyses of major and trace elements (including REE) have been performed by ICP-MS at CRPG (Nancy, France). For other samples, major and trace element (without REE) concentrations have been done by XRF spectrometry using Rigaku RIX 3000 at the Faculty of Science, Niigata University. They were determined from fused glass beads prepared with both anhydrous lithium tetraborate (Li₂B₄O₇) and lithium metaborate (LiBO₂) flux, with a sample to flux ratio of 1:2. A maximum error of 5% is estimated for the Rb/Sr ratio, based on the reproducibility of the data for the Geological Survey of Japan standard JB-1a.

All the isotope analyses were carried out using a Finigan MAT 262 thermal ionization mass spectrometer equipped with nine Faraday cup collectors at the Faculty of Science, Niigata University. Rb-Sr and Sm-Nd extraction and mass spectrometric analyses were made following the procedures described by Kagami *et al.* (1987).

Measured ⁸⁷Sr/⁸⁶Sr and ¹⁴³Nd/¹⁴⁴Nd ratios were normalized to ⁸⁶Sr/⁸⁸Sr=0.1194 and ¹⁴⁶Nd/¹⁴⁴Nd=0.7219 respectively. Whole procedure blanks were 0.55 ng Sr, 0.04 ng Sm and 0.40 ng Nd, and then their influence on the isotope ratios of measured samples are considered to be negligible. Rb-Sr concentrations are those of XRF analyses. Sm-Nd concentrations were measured using ¹⁴⁹Sm/¹⁵⁰Nd mixed spikes.

The 87 Sr/ 86 Sr ratios for this study are reported relative to 0.710242 for the NBS987 (Kagami *et al.*, 1987) and 143 Nd/ 144 Nd ratios relative to GSJ standard JNdi-1= 0.512116 that is consistent with 0.511858 of LaJolla (Tanaka *et al.*, 2000). The 143 Nd/ 144 Nd ratios for the JNdi-1 standard were measured six times and varied from 0.512080 \pm 0.000012 (2 σ) to 0.512094 \pm 0.000012 (2 σ).

Rb-Sr isochron ages are calculated using a personal computer program of Kawano (1994) which is based on the equation of York (1966) with the following decay constants: $\lambda(^{87}\text{Rb}) = 1.42 \times 10^{-11} \text{y}^{-1}$ (Steiger and Jäger, 1977) and $\lambda(^{147}\text{Sm}) = 6.54 \times 10^{-12} \text{y}^{-1}$ (Lugmair and Marti, 1978). Standard errors used for $^{87}\text{Sr}/^{86}\text{Sr}$, $^{87}\text{Rb}/^{86}\text{Sr}$, $^{147}\text{Sm}/^{144}\text{Nd}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ ratios calculation are 0.01% (1σ), 5% (1σ), 0.1% (1σ), and 0.01% (1σ) respectively.

Model ages and $\varepsilon_{\rm Nd}$ values were calculated using values of present-day depleted mantle defined by Goldstein *et al.* (1984): ¹⁴³Nd/¹⁴⁴Nd (present) = 0.51315 and ¹⁴⁷Sm/¹⁴⁴Nd (present) = 0.2136. The following CHUR parameters were used for calculations of initial $\varepsilon_{\rm Sr}$ and initial $\varepsilon_{\rm Nd}$ values: ⁸⁷Sr/⁸⁶Sr (present) = 0.7045, ⁸⁷Rb/⁸⁶Sr (present) = 0.0827, ¹⁴³Nd/¹⁴⁴Nd (present) = 0.512638 and ¹⁴⁷Sm/¹⁴⁴Nd (present) = 0.1966 (Goldstein *et al.*, 1984).

4.2. Major elements

Major elements data for some representative samples of gneisses and granitoids of

Table 1. Representative analyses of gneisses and granitoids from the Bantoum area. Analyses performed by ICP-MS (samples with REE data) or XRF (samples without REE data).

| | Biotite-g | ranites I | - (| <i>T</i> | | | | | | |
|------------------------------------|---------------|------------|-------|--|--------|---|---------------|--------|------------------------------------|-------|
| | bionie-g | granites i | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 |
| Sample | B462 | S4.5b | S4.3 | S4.5 | B46 | S3.8 | S2.5 | S2.3 | B24 | ВОО |
| • | | | | | | 70.40 | 32.3 71.11 | 71.23 | 71.34 | 72.59 |
| SiO ₂ wt% | 64.92 0.84 | 68.27 | 69.99 | 68.41 0.60 | 68.71 | | 0.50 | 0.54 | | |
| TiO ₂ | | 0.58 | 0.41 | | 0.60 | 0.47 | | | 0.48 | 0.23 |
| Al ₂ O ₃ | 16.16 | 15.07 | 14.39 | 15.34 | 15.54 | 14.45 | 13.78 | 13.99 | 14.03 | 13.91 |
| Fe ₂ O ₃ * | 4.92 | 2.87 | 2.27 | 3.20 | 3.17 | 2.96 | 3.44 | 3.52 | 2.98 | 2.36 |
| MnO | 0.05 | 0.03 | 0.02 | <l.d.< td=""><td>0.03</td><td>0.03</td><td>0.04</td><td>0.03</td><td><l.d.< td=""><td>0.02</td></l.d.<></td></l.d.<> | 0.03 | 0.03 | 0.04 | 0.03 | <l.d.< td=""><td>0.02</td></l.d.<> | 0.02 |
| MgO | 1.04 | 0.72 | 0.51 | 0.81 | 0.80 | 0.63 | 0.73 | 0.76 | 0.65 | 0.18 |
| CaO | 2.42 | 1.53 | 1.33 | 1.65 | 1.50 | 1.42 | 1.38 | 1.57 | 1.53 | 1.06 |
| Na ₂ O | 3.47 | 3.06 | 2.88 | 3.02 | 3.07 | 2.71 | 2.92 | 2.92 | 2.86 | 3.29 |
| K ₂ O | 4.92 | 6.15 | 6.08 | 6.02 | 6.02 | 5.69 | 5.09 | 4.72 | 5.33 | 5.58 |
| P_2O_5 | 0.29 | 0.27 | 0.16 | 0.21 | 0.25 | 0.16 | 0.20 | 0.20 | 0.16 | 0.14 |
| L.O.I. | 0.52 | 0.76 | 0.72 | 0.73 | 0.56 | 0.46 | 0.61 | 0.60 | 0.51 | 0.30 |
| Total | 99.55 | 99.31 | 98.75 | 99.99 | 100.25 | 99.38 | 99.80 | 100.08 | 99.87 | 99.66 |
| K ₂ O/Na ₂ O | 1.42 | 2.01 | 2.11 | 1.99 | 1.96 | 2.10 | 1.74 | 1.62 | 1.86 | 1.70 |
| Na_2O+K_2O | 8.39 | 9.21 | 8.96 | 9.04 | 9.09 | 8.40 | 8.01 | 7.64 | 8.19 | 8.87 |
| T°C Zr | 828 | 825 | 808 | 835 | 839 | 813 | 823 | 828 | 812 | 817 |
| Ba ppm | 1126 | 1361 | 1000 | 1495 | 1251 | 830 | 715 | 600 | 781 | 424 |
| Cr | 8.5 | 8 | 13 | 6.9 | 6.8 | <l.d.< td=""><td>5.3</td><td>5.8</td><td><l.d.< td=""><td>1.0</td></l.d.<></td></l.d.<> | 5.3 | 5.8 | <l.d.< td=""><td>1.0</td></l.d.<> | 1.0 |
| Ga | 26.8 | - | - | 26.9 | 25.4 | 22.7 | 24.3 | 26.4 | 23.3 | - |
| Hf | 7.67 | - | - | 11.4 | 10.3 | 7.53 | 7.62 | 8.05 | 6.49 | - |
| Nb | 29 | 15 | 17 | 17 | 16 | 19 | 23 | 29 | 18 | 38 |
| Ni | 5.9 | 8 | 8 | 10.2 | 8.6 | <l.d.< td=""><td>8.1</td><td>6.2</td><td><l.d.< td=""><td>17</td></l.d.<></td></l.d.<> | 8.1 | 6.2 | <l.d.< td=""><td>17</td></l.d.<> | 17 |
| Rb | 199 | 216 | 232 | 217 | 215 | 202 | 180 | 183 | 199 | 246 |
| Sr | 281 | 266 | 205 | 244 | 241 | 212 | 182 | 180 | 191 | 102 |
| Ta | 2.00 | - | - | 0.91 | 0.63 | 1.24 | 0.83 | 0.78 | 0.94 | - |
| Th | 29.9 | - | - | 75.5 | 72.7 | 32.6 | 54.8 | 58.4 | 41.3 | - |
| V | 63.3 | 42 | 36 | 49.0 | 47.2 | 30.2 | 34.6 | 37.6 | 30.8 | 11 |
| Y | 23.1 | 24.5 | 101 | 29.2 | 9.5 | 17.4 | 26.9 | 83.1 | 16.98 | 55 |
| Zn | 105 | 89 | - | 110 | 95.2 | 68.8 | 88.3 | 93.0 | 73.8 | 67 |
| Zr | 371 | 352 | 281 | 383 | 381 | 275 | 298 | 309 | 277 | 290 |
| La ppm | 98.3 | _ | - | 179 | 149 | 71.6 | 115 | 120 | 92.9 | - |
| Ce | 190 | _ | _ | 353 | 307 | 133 | 160 | 203 | 176 | _ |
| Pr | 21.2 | _ | _ | 41.1 | 34.6 | 15.3 | 21.5 | 22.7 | 18.9 | _ |
| Nd | 76.0 | - | _ | 145 | 124 | 51.3 | 74.6 | 79.7 | 64.6 | _ |
| Sm | 12.1 | - | _ | 19.1 | 15.7 | 7.78 | 10.5 | 12.2 | 9.32 | _ |
| Eu | 1.65 | _ | _ | 2.08 | 1.83 | 1.34 | 1.45 | 1.43 | 1.33 | _ |
| Gd | 8.27 | _ | _ | 8.82 | 7.45 | 5.23 | 6.41 | 11.1 | 5.08 | _ |
| Tb | 1.00 | _ | _ | 0.96 | 0.71 | 0.64 | 0.86 | 1.57 | 0.66 | _ |
| Dy | 4.99 | _ | _ | 3.72 | 2.60 | 3.26 | 4.00 | 9.11 | 3.05 | _ |
| Но | 0.79 | - | _ | 0.59 | 0.28 | 0.57 | 0.70 | 1.91 | 0.53 | _ |
| Er | 2.08 | - | _ | 2.12 | 0.88 | 1.66 | 1.91 | 5.63 | 1.48 | _ |
| Tm | 0.24 | _ | _ | 0.35 | 0.08 | 0.21 | 0.23 | 0.80 | 0.22 | _ |
| Yb | 1.46 | _ | _ | 2.05 | 0.47 | 1.45 | 1.27 | 4.97 | 1.16 | _ |
| Lu | 0.19 | - | _ | 0.33 | 0.08 | 0.21 | 0.22 | 0.81 | 0.20 | - |
| ьu | 0.17 | _ | - | 0.33 | 0.00 | 0.21 | 0.22 | 0.01 | 0.20 | |

Table 1 (Continued).

| | Biotite- | granites I | Ī | Leucogr | anites | | | | Granitic | dikes | |
|------------------------------------|--|------------|--|--|--|--|-------------|-----------|---|---|-----------------------|
| | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 | 19 | 20 | 21 |
| Sample | S1.1 | S3.3b | S3.3b2 | B151 | B152 | B15 | S1.9a | S1.9b | S1.42 | S1.4 | S14n |
| SiO ₂ wt% | 75.62 | 76.62 | 77.14 | 73.96 | 74.08 | 74.35 | 74.14 | 73.86 | 72.79 | 72.89 | 73.54 |
| TiO_2 | 0.22 | 0.12 | 0.16 | 0.15 | 0.15 | 0.16 | 0.11 | 0.18 | 0.24 | 0.23 | 0.24 |
| Al_2O_3 | 11.88 | 11.85 | 11.43 | 14.10 | 14.38 | 14.08 | 14.41 | 14.68 | 14.27 | 14.37 | 14.26 |
| Fe ₂ O ₃ * | 2.81 | 1.85 | 2.68 | 1.61 | 1.47 | 1.46 | 1.03 | 1.94 | 1.81 | 1.77 | 1.52 |
| MnO | 0.04 | 0.03 | 0.03 | <l.d.< td=""><td><l.d.< td=""><td>0.02</td><td>0.01</td><td>0.03</td><td><l.d.< td=""><td><l.d.< td=""><td>0.01</td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td>0.02</td><td>0.01</td><td>0.03</td><td><l.d.< td=""><td><l.d.< td=""><td>0.01</td></l.d.<></td></l.d.<></td></l.d.<> | 0.02 | 0.01 | 0.03 | <l.d.< td=""><td><l.d.< td=""><td>0.01</td></l.d.<></td></l.d.<> | <l.d.< td=""><td>0.01</td></l.d.<> | 0.01 |
| MgO | 0.16 | 0.00 | <l.d.< td=""><td>0.17</td><td>0.18</td><td>0.14</td><td>0.07</td><td>0.16</td><td>0.29</td><td>0.25</td><td>0.27</td></l.d.<> | 0.17 | 0.18 | 0.14 | 0.07 | 0.16 | 0.29 | 0.25 | 0.27 |
| CaO | 0.78 | 0.67 | 0.78 | 0.83 | 0.89 | 0.63 | 0.63 | 0.72 | 0.71 | 0.71 | 0.72 |
| Na ₂ O | 2.42 | 2.21 | 2.15 | 3.04 | 3.22 | 2.78 | 3.12 | 2.74 | 3.00 | 3.05 | 3.19 |
| K_2O | 5.35 | 6.01 | 5.49 | 5.41 | 5.38 | 6.04 | 5.52 | 4.78 | 6.34 | 6.44 | 6.28 |
| P_2O_5 | 0.05 | 0.03 | <l.d.< td=""><td>0.06</td><td>0.09</td><td>0.08</td><td>0.08</td><td>0.13</td><td>0.14</td><td>0.09</td><td>0.14</td></l.d.<> | 0.06 | 0.09 | 0.08 | 0.08 | 0.13 | 0.14 | 0.09 | 0.14 |
| L.O.I. | 0.50 | O,55 | 0.44 | 0.44 | 0.58 | 0.66 | 0.89 | 1.01 | 0.57 | 0.46 | 0.56 |
| Total | 99.83 | 99.94 | 100.30 | 99.77 | 100.42 | 100.42 | 100.01 | 100.23 | 100.16 | 100.26 | 100.72 |
| K ₂ O/Na ₂ O | 2.21 | 2.72 | 2.55 | 1.78 | 1.67 | 2.17 | 1.77 | 1.74 | 2.11 | 2.11 | 1.97 |
| Na ₂ O+K ₂ O | 7.77 | 8.22 | 7.64 | 8.45 | 8.60 | 8.83 | 8.64 | 7.52 | 9.34 | 9.49 | 9.47 |
| T°C Zr | 866 | 846 | 851 | 742 | 745 | 740 | 719 | 736 | 753 | 751 | 752 |
| Ba ppm | 372 | 119 | 123 | 127 | 145 | 175 | 112 | 63.5 | 380 | 372 | 4070 |
| Cr | <l.d.< td=""><td>7.0</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td>8</td><td>0</td><td>6.7</td><td><l.d.< td=""><td><l.d.< td=""><td>9</td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | 7.0 | <l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td>8</td><td>0</td><td>6.7</td><td><l.d.< td=""><td><l.d.< td=""><td>9</td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td><l.d.< td=""><td>8</td><td>0</td><td>6.7</td><td><l.d.< td=""><td><l.d.< td=""><td>9</td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td>8</td><td>0</td><td>6.7</td><td><l.d.< td=""><td><l.d.< td=""><td>9</td></l.d.<></td></l.d.<></td></l.d.<> | 8 | 0 | 6.7 | <l.d.< td=""><td><l.d.< td=""><td>9</td></l.d.<></td></l.d.<> | <l.d.< td=""><td>9</td></l.d.<> | 9 |
| Ga | 23.6 | - | 25.9 | 29.4 | 31.0 | - | - | - | 21.9 | 22.0 | - |
| Hf | 12.0 | - | 14.08 | 3.55 | 3.91 | - | - | - | 4.01 | 4.04 | - |
| Nb | 29 | 22 | 24 | 21 | 21 | 21 | 24 | 36 | 10 | 11 | 12 |
| Ni | <l.d.< td=""><td>2</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td>6</td><td>13</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | 2 | <l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td>6</td><td>13</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td>6</td><td>13</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td><l.d.< td=""><td>6</td><td>13</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td>6</td><td>13</td><td><l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | 6 | 13 | <l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td><l.d.< td=""></l.d.<></td></l.d.<> | <l.d.< td=""></l.d.<> |
| Rb | 135 | 158 | 132 | 346 | 367 | 370 | 386 | 390 | 237 | 244 | 257 |
| Sr | 60.1 | 30.0 | 24.8 | 35.6 | 42.0 | 45.5 | 33.0 | 21.5 | 87.2 | 88.3 | 97.3 |
| Та | 2.19 | - | 0.49 | 1.19 | 1.26 | - | - | - | 0.50 | 0.49 | - |
| Th | 14.7 | - | 21.7 | 33.6 | 32.5 | - | - | - | 31.7 | 31.7 | - |
| V | 8.4 | 7 | <l.d.< td=""><td><l.d.< td=""><td><l.d.< td=""><td>4</td><td>7</td><td>7</td><td>4.7</td><td>4.7</td><td>7</td></l.d.<></td></l.d.<></td></l.d.<> | <l.d.< td=""><td><l.d.< td=""><td>4</td><td>7</td><td>7</td><td>4.7</td><td>4.7</td><td>7</td></l.d.<></td></l.d.<> | <l.d.< td=""><td>4</td><td>7</td><td>7</td><td>4.7</td><td>4.7</td><td>7</td></l.d.<> | 4 | 7 | 7 | 4.7 | 4.7 | 7 |
| Y | 108 | 69 | 81.3 | 16.3 | 16.2 | 13 | 19 | 22 | 33.2 | 32.4 | 43 |
| Zn | 70.9 446 | 51 386 | 65.6 393 | 79.8 | 78.8 109 | 102 | 55 74 | 108 79 | 61.1 | 64.1 | 122 |
| Zr | 446 | 380 | 393 | 104 | 109 | 103 | - | /9 - | 130 | 135 | 133 |
| La ppm | 96.2 | _ | 124 | 34.8 | 32.5 | _ | _ | _ | 52.7 | 50.3 | _ |
| Ce | 207 | - | 247 | 71.2 | 65.4 | _ | _ | _ | 88.9 | 86.1 | _ |
| Pr | 24.5 | - | 29.4 | 8.35 | 7.64 | - | - | _ | 10.9 | 10.6 | _ |
| Nd | 94.6 | - | 113 | 29.6 | 28.0 | - | - | - | 40.8 | 39.2 | - |
| Sm | 21.6 | - | 24.4 | 6.87 | 6.34 | - | - | - | 8.99 | 8.77 | - |
| Eu | 1.35 | - | 0.96 | 0.42 | 0.48 | ~ | - | - | 0.85 | 0.88 | - |
| Gd | 21.4 | - | 21.6 | 4.75 | 4.76 | - | - | - | 8.45 | 8.45 | - |
| Tb | 3.33 | - | 3.31 | 0.69 | 0.65 | - | - | - | 1.12 | 1.06 | - |
| Dy | 19.5 | - | 18.2 | 2.92 | 2.87 | - | - | - | 5.17 | 5.17 | - |
| Но | 3.83 | - | 3.24 | 0.49 | 0.49 | - | - | - | 1.02 | 0.94 | - |
| Er | 9.98 | - | 7.58 | 1.17 | 1.06 | - | - | - | 2.54 | 2.15 | - |
| Tm | 1.34 | - | 0.92 | 0.12 | 0.12 | - | - | - | 0.30 | 0.28 | - |
| Yb | 7.96 | - | 5.03 | 0.69 | 0.71 | - | - | - | 1.66 | 1.63 | - |
| Lu | 1.00 | - | 0.71 | 0.11 | 0.08 | - | - | - | 0.26 | 0.25 | - |

Table 1 (Continued).

| ###################################### | Quartz-m | onzonite | s | | | Plagiocla | se-rich g | neiss | |
|--|----------|----------|-------|--------|-------|-----------|-----------|-------|-------|
| | 22 | 23 | 24 | 25 | 26 | 27 | 28 | 29 | 30 |
| Sample | B47 | S4.7 | S4.6 | S3.6 | S3.7c | S4.1 | S4.1a | S4.1j | B43 |
| SiO ₂ wt% | 54.29 | 55.18 | 55.17 | 57.41 | 57.72 | 63.10 | 66.99 | 62.39 | 65.99 |
| TiO ₂ | 1.89 | 1.84 | 1.79 | 1.46 | 1.39 | 1.16 | 1.13 | 1.38 | 1.16 |
| Al_2O_3 | 16.63 | 15.90 | 15.77 | 16.18 | 16.13 | 14.47 | 13.78 | 14.80 | 14.14 |
| Fe ₂ O ₃ * | 9.09 | 8.32 | 8.23 | 8.06 | 7.84 | 7.28 | 7.13 | 8.88 | 7.37 |
| MnO | 0.12 | 0.11 | 0.10 | 0.10 | 0.11 | 0.14 | 0.15 | 0.15 | 0.16 |
| MgO | 3.26 | 3.02 | 3.26 | 3.52 | 3.61 | 2.63 | 1.18 | 2.04 | 1.32 |
| CaO | 5.48 | 5.09 | 5.10 | 5.22 | 5.04 | 5.01 | 3.29 | 4.58 | 3.53 |
| Na ₂ O | 3.34 | 3.19 | 3.35 | 3.19 | 3.29 | 3.29 | 3.16 | 3.30 | 3.24 |
| K ₂ O | 4.38 | 4.51 | 4.34 | 3.58 | 3.70 | 1.85 | 2.14 | 1.99 | 2.08 |
| P_2O_5 | 0.95 | 1.10 | 1.07 | 0.48 | 0.47 | 0.30 | 0.33 | 0.29 | 0.34 |
| L.O.I. | 0.66 | 0.69 | 0.61 | 0.81 | 0.58 | 0.51 | 0.30 | 0.12 | 0.26 |
| Total | 100.09 | 98.94 | 98.79 | 100.01 | 99.88 | 99.74 | 99.58 | 99.92 | 99.58 |
| K ₂ O/Na ₂ O | 1.31 | 1.41 | 1.30 | 1.12 | 1.12 | 0.56 | 0.68 | 0.60 | 0.64 |
| Na ₂ O+K ₂ O | 7.72 | 7.70 | 7.69 | 6.77 | 6.99 | 5.14 | 5.30 | 5.29 | 5.32 |
| T°C Zr | 753 | 760 | 763 | 768 | 757 | | | | |
| Ba ppm | 2020 | 2134 | 2393 | 1493 | 1269 | 646 | 1198 | 746 | 1090 |
| Cr | 39.6 | 45 | 55 | 98.7 | 91.6 | 62 | 7 | 53.3 | 16 |
| Ga | 24.2 | - | - | 22.9 | 22.6 | - | - | 25.7 | - |
| Hf | 8.28 | - | - | 7.91 | 6.92 | - | - | 7.71 | - |
| Nb | 28 | 31 | 23 | 25 | 24 | 18 | 20 | 20 | 21 |
| Ni | 26.6 | 22 | 37 | 47.0 | 48.7 | 24 | 9 | 28.0 | 8 |
| Rb | 173 | 171 | 161 | 196 | 148 | 54 | 75 | 62 | 74 |
| Sr | 1102 | 1042 | 1069 | 609 | 548 | 237 | 297 | 290 | 306 |
| Ta | 1.74 | - | - | 1.53 | 1.65 | - | - | 1.53 | - |
| Th | 17.0 | - | - | 18.4 | 17.4 | - | - | 9.71 | - |
| V | 153 | 169 | 146 | 133 | 122 | 103 | 39 | 101 | 59 |
| Y | 26.1 | 27 | 28 | 20.5 | 18.5 | 54 | 55 | 52.7 | 54 |
| Zn | 139 | - | 106 | 114 | 112 | 88 | 79 | 117 | - |
| Zr | 346 | 357 | 378 | 324 | 282 | 329 | 426 | 333 | 392 |
| La ppm | 107 | - | - | 96.3 | 79.7 | - | - | 18.3 | - |
| Ce | 211 | - | - | 192 | 145 | - | - | 79.1 | - |
| Pr | 23.7 | - | - | 20.4 | 16.8 | - | - | 6.22 | - |
| Nd | 92.4 | - | - | 75.0 | 62.6 | - | - ' | 26.6 | - |
| Sm | 14.3 | - | - | 11.0 | 9.74 | - | - | 7.20 | - |
| Eu | 3.62 | - | - | 2.49 | 2.19 | - | - | 2.51 | - |
| Gd | 9.47 | - | - | 7.74 | 6.15 | - | - | 8.29 | - |
| Tb | 1.20 | - | - | 0.93 | 0.90 | - | - | 1.38 | - |
| Dy | 5.99 | - | - | 4.55 | 4.37 | - | - | 8.61 | - |
| Но | 0.89 | - | - | 0.66 | 0.68 | - | - | 1.83 | - |
| Er | 2.22 | - | - | 1.71 | 1.77 | - | - | 4.96 | - |
| Tm | 0.26 | - | - | 0.21 | 0.22 | - | - | 0.73 | - |
| Yb | 1.75 | - | - | 1.37 | 1.37 | - | - | 4.86 | - |
| Lu | 0.27 | - | - | 0.23 | 0.21 | - | - | 0.68 | - |

Table 1 (Continued).

| Sample Both Sample Sample Both Sample Both Sample Sample Sample Both Sample Sampl | | | | 1010 1 (00 | | ,. | | | |
|---|----------------------------------|--|-------|---|---|--------|-------|-------|-------|
| Sample B21 S3.5c B31 B25 S2.62 S3.1 S3.11 S3.12 SiO ₂ wt% 71.12 76.60 77.52 69.03 70.11 78.64 73.68 71.76 TiO ₂ 0.26 0.28 0.28 0.36 0.39 0.14 0.37 0.17 Al ₂ O ₃ 14.60 10.99 11.00 14.70 13.88 11.23 12.67 13.57 Fe _F O ₃ * 4.25 3.33 3.26 4.36 4.39 2.22 2.98 3.22 MnO 0.06 0.011 0.10 0.05 0.08 0.03 0.05 0.09 MgO <a block"="" href="https://doi.org/10.1001/10.</td><td></td><td></td><td></td><td>gneiss</td><td></td><td></td><td></td><td></td><td></td></tr><tr><td><math display="> \begin{array}{c ccccccccccccccccccccccccccccccccccc | | 31 | 32 | 33 | 34 | 35 | 36 | 37 | 38 |
| TiO₂ 0.26 0.28 0.28 0.36 0.39 0.14 0.37 0.17 Al₂O₃ 14.60 10.99 11.00 14.70 13.88 11.23 12.67 13.57 Fe₀O₃* 4.25 3.33 3.26 4.36 4.39 2.22 2.98 3.22 MnO 0.06 0.11 0.10 0.05 0.08 0.03 0.05 0.09 MgO 1.10 0.05 0.24 0.23 0.02 0.31 0.01 CaO 1.16 0.94 1.02 1.59 1.36 1.11 0.98 1.12 Na₂O 3.95 2.14 2.25 3.03 3.43 2.90 2.72 2.56 K₂O 4.20 4.60 4.36 5.94 5.43 3.42 5.92 6.57 P₂O₃ 0.09 0.04 0.03 0.08 0.10 0.02 0.12 0.05 Lo.I. 1.05 1.33 | Sample | B21 | S3.5c | B31 | B25 | S2.62 | S3.1 | S3.11 | S3.12 |
| $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$ | SiO ₂ wt% | 71.12 | 76.60 | 77.52 | 69.03 | 70.11 | 78.64 | 73.68 | 71.76 |
| Fe ₂ O ₃ * 4.25 3.33 3.26 4.36 4.39 2.22 2.98 3.22 MnO 0.06 0.11 0.10 0.05 0.08 0.03 0.05 0.09 MgO <l.d.< th=""> 1.10 0.05 0.24 0.23 0.02 0.31 0.01 CaO 1.16 0.94 1.02 1.59 1.36 1.11 0.98 1.12 Na₂O 3.95 2.14 2.25 3.03 3.43 2.90 2.72 2.56 K₂O 4.20 4.60 4.36 5.94 5.43 3.42 5.29 6.57 P_PO₅ 0.09 0.04 0.03 0.08 0.10 0.02 0.12 0.04 LO.I. 0.66 0.46 0.24 0.43 0.33 0.44 0.32 0.57 P₂O₅ 0.09 0.04 0.03 0.08 0.10 0.02 0.12 0.04 LO.I. 1.06 2.15 1.93 1.</l.d.<> | TiO ₂ | 0.26 | 0.28 | 0.28 | 0.36 | 0.39 | 0.14 | 0.37 | 0.17 |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$ | Al_2O_3 | 14.60 | 10.99 | 11.00 | 14.70 | 13.88 | 11.23 | 12.67 | 13.57 |
| $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$ | Fe ₂ O ₃ * | 4.25 | 3.33 | 3.26 | 4.36 | 4.39 | 2.22 | 2.98 | 3.22 |
| $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$ | MnO | 0.06 | 0.11 | 0.10 | 0.05 | 0.08 | 0.03 | 0.05 | 0.09 |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$ | MgO | <l.d.< td=""><td>1.10</td><td>0.05</td><td>0.24</td><td>0.23</td><td>0.02</td><td>0.31</td><td>0.01</td></l.d.<> | 1.10 | 0.05 | 0.24 | 0.23 | 0.02 | 0.31 | 0.01 |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$ | CaO | 1.16 | 0.94 | 1.02 | 1.59 | 1.36 | 1.11 | 0.98 | 1.12 |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$ | Na ₂ O | 3.95 | 2.14 | 2.25 | 3.03 | 3.43 | 2.90 | 2.72 | |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$ | | | | 4.36 | | | | | |
| $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$ | P ₂ O ₅ | 0.09 | 0.04 | 0.03 | | 0.10 | | | |
| $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$ | | | | | | | | | |
| $\begin{array}{c ccccccccccccccccccccccccccccccccccc$ | | | | | | | | | |
| Na ₂ O+K ₂ O T°C Zr 8.15 6.74 6.61 8.97 8.86 6.32 8.01 9.13 Ba ppm 435 877 1001 784 766 142 549 239 Cr <l.d.< th=""> 1 14 <l.d.< th=""> 19 53 4 0 Ga 39.4 - - 35.6 - - - - Hf 20.58 - - 13.30 - - - - Nb 83 40 43 68 87 19 38 60 Ni <l.d.< th=""> 16 <l.d.< th=""> <l.d.< th=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - - Th 16.2 - - 18.4 - -</l.d.<></l.d.<></l.d.<></l.d.<></l.d.<> | | | | | | | | | |
| Ba ppm 435 877 1001 784 766 142 549 239 Cr <l.d.< td=""> 1 14 <l.d.< td=""> 19 53 4 0 Ga 39.4 - - 35.6 - - - - Hf 20.58 - - 13.30 - - - - Nb 83 40 43 68 87 19 38 60 Ni <l.d.< td=""> 16 <l.d.< td=""> <l.d.< td=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - - Th 16.2 - 18.4 - - - - - <tr< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></tr<></l.d.<></l.d.<></l.d.<></l.d.<></l.d.<> | | | | | | | | | |
| Ba ppm 435 877 1001 784 766 142 549 239 Cr <l.d.< th=""> 1 14 <l.d.< th=""> 19 53 4 0 Ga 39.4 - - 35.6 - - - - Hf 20.58 - - 13.30 - - - - Nb 83 40 43 68 87 19 38 60 Ni <l.d.< th=""> 16 <l.d.< th=""> <l.d.< th=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - Th 116.2 - - 118.4 - - - - V<!--</td--><td></td><td></td><td>0</td><td>0.01</td><td>0.7.</td><td></td><td>0.02</td><td>0.01</td><td>,,,,,</td></l.d.<></l.d.<></l.d.<></l.d.<></l.d.<> | | | 0 | 0.01 | 0.7. | | 0.02 | 0.01 | ,,,,, |
| Cr | | | | | | | | | |
| Ga 39.4 - - 35.6 - - - - Hf 20.58 - - 13.30 - - - - Nb 83 40 43 68 87 19 38 60 Ni <l.d.< td=""> 16 <l.d.< td=""> <l.d.< td=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 -</l.d.<></l.d.<></l.d.<> | Ba ppm | 435 | 877 | 1001 | 784 | 766 | 142 | 549 | 239 |
| Hf 20.58 - - 13.30 - - - - Nb 83 40 43 68 87 19 38 60 Ni <l.d.< td=""> 16 <l.d.< td=""> <l.d.< td=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - - Th 116.2 - - 18.4 - - - - V <l.d.< td=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr</l.d.<></l.d.<></l.d.<></l.d.<> | Cr | <l.d.< td=""><td>1</td><td>14</td><td><l.d.< td=""><td>19</td><td>53</td><td>4</td><td>0</td></l.d.<></td></l.d.<> | 1 | 14 | <l.d.< td=""><td>19</td><td>53</td><td>4</td><td>0</td></l.d.<> | 19 | 53 | 4 | 0 |
| Nb 83 40 43 68 87 19 38 60 Ni <l.d.< th=""> 16 <l.d.< th=""> <l.d.< th=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - - Th 16.2 - - 18.4 - - - - V <l.d.< th=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La</l.d.<></l.d.<></l.d.<></l.d.<> | Ga | 39.4 | - | - | 35.6 | - | - | - | - |
| Ni <l.d.< th=""> 16 <l.d.< th=""> <l.d.< th=""> 15 8 3 12 Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 -</l.d.<></l.d.<></l.d.<> | Hf | 20.58 | - | - | 13.30 | - | - | - | - |
| Rb 241 176 145 182 170 111 166 156 Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - - Th 16.2 - - 18.4 - - - - V <ld.< td=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ze 195 - - 219 - - - - <</ld.<> | Nb | 83 | 40 | 43 | 68 | 87 | 19 | 38 | 60 |
| Sr 46.1 91 90 86.8 90 51 136 44 Ta 2.46 - - 2.02 - - - - Th 16.2 - - 18.4 - - - - V <l.d.< td=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 21.4 - - 25.2 - - - - Nd<td>Ni</td><td><l.d.< td=""><td>16</td><td><l.d.< td=""><td><l.d.< td=""><td>15</td><td>8</td><td>3</td><td>12</td></l.d.<></td></l.d.<></td></l.d.<></td></l.d.<> | Ni | <l.d.< td=""><td>16</td><td><l.d.< td=""><td><l.d.< td=""><td>15</td><td>8</td><td>3</td><td>12</td></l.d.<></td></l.d.<></td></l.d.<> | 16 | <l.d.< td=""><td><l.d.< td=""><td>15</td><td>8</td><td>3</td><td>12</td></l.d.<></td></l.d.<> | <l.d.< td=""><td>15</td><td>8</td><td>3</td><td>12</td></l.d.<> | 15 | 8 | 3 | 12 |
| Ta 2.46 - - 2.02 - - - - Th 16.2 - - 18.4 - - - V <l.d.< td=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 21.4 - - 25.2 - - - - Nd 79.3 - - 88.4 - - - - Sm 17.5</l.d.<> | Rb | 241 | 176 | 145 | 182 | 170 | 111 | 166 | 156 |
| Th 16.2 - - 18.4 - - - - V <l.d.< th=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 214 - - 25.2 - - - - Nd 79.3 - - 88.4 - - - - Sm 17.5 - 17.0 - - - - Gd 15.8</l.d.<> | Sr | 46.1 | 91 | 90 | 86.8 | 90 | 51 | 136 | 44 |
| V <l.d.< th=""> 7 5 3.5 8 7 18 7 Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Ce 195 - - 219 - - - - Pr 214 - - 25.2 - - - - Sm 17.5 - - 17.0 - - - - Sm 17.5 - 15.9 - - - - Gd 15.8</l.d.<> | Та | 2.46 | - | _ | 2.02 | - | - | - | - |
| Y 105 103 71 66.9 116 65 68 164 Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 214 - - 219 - - - - Nd 79.3 - - 25.2 - - - - Nd 79.3 - - 25.2 - - - - Sm 17.5 - - 17.0 - - - - Eu 1.42 - - 1.82 - - - - Tb | Th | 16.2 | - | _ | 18.4 | - | - | - | - |
| Zn 149 80 - 92.6 93 53 75 115 Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 214 - - 25.2 - - - - Nd 79.3 - - 25.2 - - - - Nd 79.3 - - 25.2 - - - - Sm 17.5 - - 17.0 - - - - Sm 17.5 - - 182 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy <th< td=""><td>V</td><td><l.d.< td=""><td>7</td><td>5</td><td>3.5</td><td>8</td><td>7</td><td>18</td><td>7</td></l.d.<></td></th<> | V | <l.d.< td=""><td>7</td><td>5</td><td>3.5</td><td>8</td><td>7</td><td>18</td><td>7</td></l.d.<> | 7 | 5 | 3.5 | 8 | 7 | 18 | 7 |
| Zr 840 559 572 541 562 339 488 646 La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 21.4 - - 25.2 - - - - Nd 79.3 - - 88.4 - - - - Sm 17.5 - 17.0 - - - - Sm 17.5 - 17.0 - - - - Eu 1.42 - - 1.82 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - 2.70 - - - - Er 9.37 - 6.77 - - - - Tm 1.33 - </td <td>Y</td> <td>105</td> <td>103</td> <td>71</td> <td>66.9</td> <td>116</td> <td>65</td> <td>68</td> <td>164</td> | Y | 105 | 103 | 71 | 66.9 | 116 | 65 | 68 | 164 |
| La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 214 - - 25.2 - - - - Nd 79.3 - - 88.4 - - - - Sm 17.5 - - 17.0 - - - - Eu 1.42 - - 1.82 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - 2.70 - - - - Er 9.37 - 6.77 - - - - Tm 1.33 - 0.82 | Zn | 149 | 80 | _ | 92.6 | 93 | 53 | 75 | 115 |
| La ppm 87.6 - - 113 - - - - Ce 195 - - 219 - - - - Pr 214 - - 25.2 - - - - Nd 79.3 - - 88.4 - - - - Sm 17.5 - - 17.0 - - - - Eu 1.42 - - 1.82 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - 2.70 - - - - Er 9.37 - 6.77 - - - - Tm 1.33 - 0.82 | Zr | 840 | 559 | 572 | 541 | 562 | 339 | 488 | 646 |
| Ce 195 - 219 - <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> | | | | | | | | | |
| Pr 2 1.4 - - 25.2 - - - Nd 79.3 - - 88.4 - - - Sm 17.5 - - 17.0 - - - Eu 1.42 - - 1.82 - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - - 0.82 - - - - Yb 8.03 - - 4.78 - - - - - | La ppm | 87.6 | - | - | 113 | - | - | - | - |
| Nd 79.3 - - 88.4 - - - - Sm 17.5 - - 17.0 - - - - Eu 1.42 - - 1.82 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - 0.82 - - - - - Yb 8.03 - 4.78 - - - - - | Ce | 195 | - | - | 219 | - | - | - | - |
| Sm 17.5 - - 17.0 - - - - Eu 1.42 - - 1.82 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - - 0.82 - - - - Yb 8.03 - - 4.78 - - - - - | Pr | 2 1.4 | - | - | 25.2 | - | - | - | - |
| Eu 1.42 - - 1.82 - - - - Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - - 0.82 - - - - Yb 8.03 - - 4.78 - - - - - | Nd | 79.3 | - | - | 88.4 | - | - | - | - |
| Gd 15.8 - - 15.9 - - - - Tb 2.80 - - 2.37 - - - - Dy 17.5 - - 13.9 - - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - - 0.82 - - - - Yb 8.03 - - 4.78 - - - - - | Sm | 17.5 | - | - | 17.0 | - | - | - | - |
| Tb 2.80 - - 2.37 - - - Dy 17.5 - - 13.9 - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - - 0.82 - - - - Yb 8.03 - - 4.78 - - - - | Eu | 1.42 | - | - | 1.82 | - | - | - | - |
| Dy 17.5 - - 13.9 - - - - Ho 3.59 - - 2.70 - - - - Er 9.37 - - 6.77 - - - - Tm 1.33 - - 0.82 - - - - Yb 8.03 - - 4.78 - - - - | Gd | 15.8 | - | - | 15.9 | - | - | - | - |
| Ho 3.59 2.70 | Tb | 2.80 | - | - | 2.37 | - | - | - | - |
| Ho 3.59 2.70 | Dy | 17.5 | - | _ | 13.9 | - | - | - | - |
| Er 9.37 6.77 | - | | - | - | | - | - | - | - |
| Tm 1.33 0.82 Yb 8.03 4.78 | Er | | - | - | | - | _ | _ | - |
| Yb 8.03 4.78 | | | _ | _ | | - | _ | - | _ |
| | | | _ | _ | | - | _ | _ | _ |
| | | | | | | _ | _ | _ | _ |

^{*} total iron as Fe₂O₃; L.O.I. loss on ignition; <L.D. below the limit of detection; - not analyzed

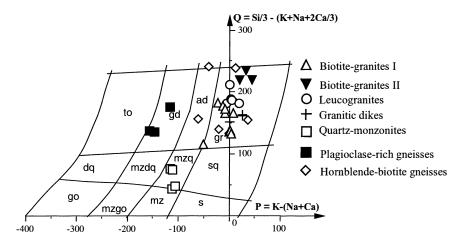


Fig. 5. Position of the Bantoum granitoids and gneisses in the P=Si/3-(K+Na+2Ca/3) vs P=K-(Na+Ca) (gram-atoms×10³) multicationic nomenclature diagram of igneous rocks after Debon and Le Fort (1983). gr, granite; ad, adamellite; gd, granodiorite; to, tonalite; sq, quartz-syenite; mzq, quartz-monzonite; mzdq, quartz-monzodiorite; dq, quartz-diorite; s, syenite; mz, monzonite; mzgo, monzodiorite; go, gabbro.

the Bantoum area are listed in Table 1. We used the P = K - (Na + Ca) vs. Q = Si/3 -(K+Na+2Ca/3) multi-cationic diagram of Debon and Le Fort (1983) for nomenclature chemical composition of the different petrographic types, as this diagram takes into account the main major elements. Biotite-granites, granitic dike, leucogranites and hornblende-biotite gneisses share the granitic field (Fig. 5); plagioclase-rich gneiss has a granodioritic composition. SiO₂ contents range from 56-58 wt% in quartz-monzonites and 65 to 77 wt% for other granites; the gap between 56-65 wt% silica content is noticeable. In the Harker diagrams (Fig. 6), the most felsic samples (75–77 wt% SiO₂) amongst biotite-granites show some slight chemical differences with other terms, in particular their much lower content in MgO, CaO, TiO₂, Al₂O₃ and Na₂O. Hereafter we refer to the less felsic biotite-granites as "granites I" and more felsic ones as "granites II", even though they show no significant difference on the field. All granitoids have high total alkali contents (6.8% < Na₂O + K₂O < 9.5%). They are hyperpotassic (K₂O =3.6-6.4 wt%), having K_2O/Na_2O ratios ranging from 1.1 to 2.7 with lowest values in quartz-monzonites (1.1-1.4). High TiO₂ contents (1.46-1.89 wt%) in quartzmonzonites could be explained by the abundance of sphene (up to 3 vol% in some samples). Regarding the Al saturation index (A/CNK = molar Al₂O₃/(CaO + Na₂O + (K₂O₁), quartz-monzonites are metaluminous while biotite-granites and granitic dikes are slightly peraluminous and leucogranites are peraluminous. In the A/CNK vs SiO₂ diagram (Fig. 7), only leucogranites plot in the field of S-type granites and other granites are in the field of I-type granitoids (Chappell and White, 1992). In the (Na₂O +K₂O) vs SiO₂ and AFM diagrams (Fig. not shown) they show a calc-alkaline trend. In addition the presence of normative corundum in leucogranites (1.9-4%) and biotite-granites (0.6-1.3%) and their absence in quartz-monzonites supports this subdivision.

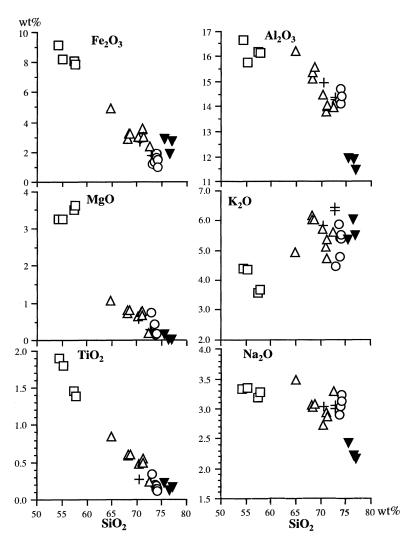


Fig. 6. Harker variation diagrams for some selected major elements. Symbols as in Fig. 5.

4.3. Trace elements

The trace element contents of Bantoum granitoids and gneisses are reported in Table 1. Although they are in general scattered on variation diagrams (Fig. 8), some features are noteworthy. Ba abundance is high in quartz-monzonites (1268–2393 ppm), moderate in granites I and dikes (371–1495 ppm) and low in granites II and leucogranites (63–372 ppm). Ba/Sr ratios range between 1.8–2.5 in quartz-monzonites and are alike in other groups (3–6). Rb/Ba < 1 except in leucogranites (2.5–6.1); likewise Rb/Sr ratios are much higher in leucogranites (8.7–18.1) than in biotite-granites and granitic dikes (0.7–2.8) and quartz-monzonites (<0.4). K/Rb ratios are also significantly lower in leucogranites (126–156) compared to quartz-monzonites and

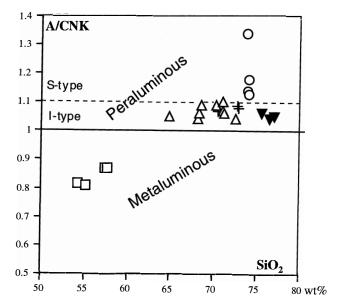


Fig. 7. Molar $Al_2O_3/(CaO+Na_2O+K_2O)$ vs SiO_2 diagram for the Bantoum granitoids. Boundary between I-type and S-type according to Chappell and White (1992). Symbols as in Fig. 5.

biotite-granites (247-416); these values are similar to those observed in continental calc-alkaline igneous suites (e.g. Bertrand et al., 1984; Ayuso and Arth, 1992).

Trace elements distribution patterns show that all these rocks have a distinctive depletion in Nb, Ta, Sr and Ti (except quartz-monzonites) relative to other trace elements, and are enriched in LILE (large ion lithophile elements) compared to anomalies in Sr, Ba and Ti; and distinctively higher contents of Th are observed in the biotite-granites I. Quartz-monzonites display less pronounced negative anomalies in Sr and Ti, and lower Rb, Y and Yb values resulting in more "fractionated" trace element distribution pattern, characteristic of calc-alkaline arc granitoids. Also noticeable is the similarity between trace elements distribution patterns of biotite-granites II and hornblende-biotite gneisses (Fig. 9d). Overall, the enrichment in the incompatible elements (K, Rb, Ba) and depletion in high field strength elements (HFSE) supports the importance of the crustal source component in the genesis of the Bantoum granitoids.

The conventional tectonic discrimination diagrams of Pearce *et al.* (1984; Fig. 10) are used to summarize some of the salient trace element geochemical features of the Bantoum gneisses and granites. Leucogranites have trace element compositions which straddle the field boundary between volcanic arc granites and syn-collisional granites. High Rb/Zr ratios (3–5) and lower K/Rb ratios of the leucogranites support their syn-collisional characters (Harris *et al.*, 1986). Quartz-monzonites, granites I and granitic dikes plot within the field of volcanic arc granitoids, whereas "granites II" and both hornblende-biotite and plagioclase-rich gneisses plot within the within plate granitoids field.

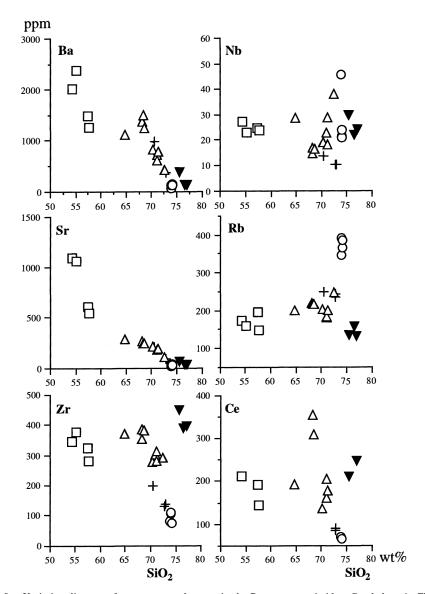


Fig. 8. Variation diagrams for some trace elements in the Bantoum granitoids. Symbols as in Fig. 5.

4.4. Rare earth elements

Total REE is lower (151–224 ppm) in leucogranites and granitic dikes comparatively to both biotite-granites I and II and quartz-monzonites (294–757 ppm). Chondrite-normalized REE patterns (Fig. 11) in general indicate strong LREE enrichment but variable LREE/HREE contents ($\text{La}_N/\text{Yb}_N=8$ –61). Quartz-monzonites and biotite-granites I display similar consistent fractionation trend within the LREE and HREE groups ($\text{La}_N/\text{Yb}_N=33$ –61; $\text{La}_N/\text{Sm}_N=5$ –7 and $\text{Gd}_N/\text{Yb}_N=3$ –5; $\text{Ce}_N/\text{Yb}_N=23$ –44), except for two samples (B46 and S2.3) of granites I that are significantly much more

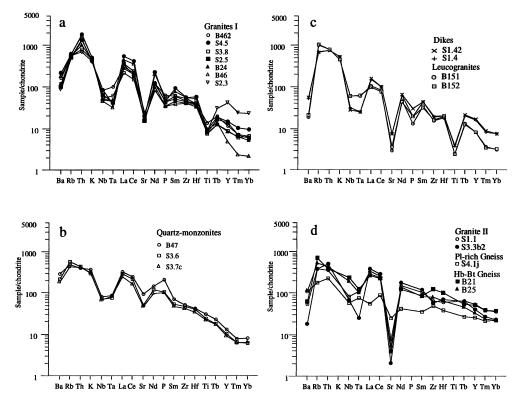


Fig. 9. Chondrite-normalized trace-element distribution for the Bantoum granitoids and gneisses.

Normalization values after Thompson et al. (1984).

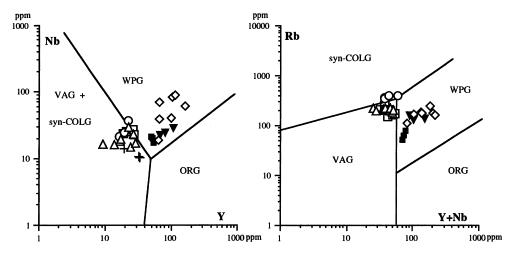


Fig. 10. Nb vs Y and Rb vs Y+Nb discrimination diagrams (Pearce et al., 1984) for the granites and gneisses of Bantoum area. WPG, Within-Plate Granites; VAG, Volcanic Arc Granites; syn-COLG, syn-Collisional Granites; ORG, Oceanic Ridge Granites. Symbols as in Fig. 5.

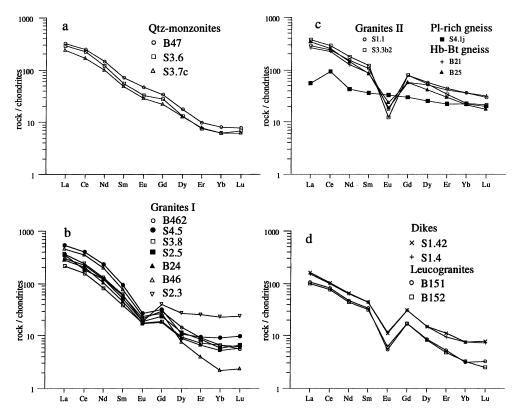


Fig. 11. Chondrite-normalized rare earth element patterns for the Bantoum granitoids and amphibole-biotite gneisses. Normalization values after Nakamura (1974).

fractionated in HREE ($Gd_N/Yb_N=13$ and $La_N/Yb_N=210$) and less fractionated ($Gd_N/Yb_N=2$ and $La_N/Yb_N=16$) respectively (Fig. 11b). The difference between granites II and granites I is that the former are more enriched in HREE ($Gd_N/Yb_N=2-3$) and show less fractionate trends ($La_N/Yb_N=8-17$); like for trace elements distribution patterns these granites II and hornblende-biotite gneisses display quite similar REE patterns (Fig. 11c). Negative Eu anomalies are not significant in quartz-monzonites whereas they are more consistent in granites I ($Eu/Eu^*=0.49-0.55$) and pronounced in granites II ($Eu/Eu^*=0.13-0.19$), granitic dikes ($Eu/Eu^*=0.13-0.19$), and leucogranites ($Eu/Eu^*=0.22-0.27$). The chondrite-normalized REE pattern for the plagio-clase-rich gneiss sample S4.1j is quite flat (Fig. 11c) showing no fractionation.

4.5. Geothermometry

In the ternary Qtz-Ab-Or normative diagram (Fig. 12a), the majority of biotite-granite I, granitic dikes and leucogranites compositions plot close to the granitic minimum determined experimentally at $P_{\rm H_2O}=2\,\rm kb$ and water-undersaturated ($a_{\rm H_2O}=0.5$) by Holtz *et al.* (1992). The relatively high normative Or content of the natural granites may be explained by pronounced H₂O-undersaturated conditions in the source materials of these granites as for the examples observed for Harney Peak granites

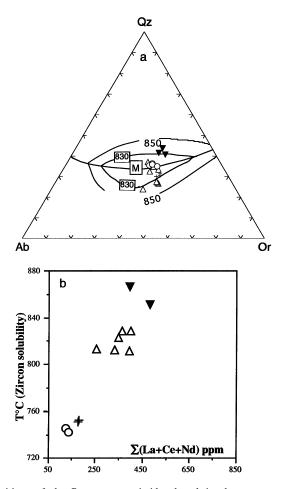


Fig. 12. a) Compositions of the Bantoum granitoids plotted in the ternary Qtz-Ab-Or normative diagram determined experimentally by Holtz et al., (1992) (at P_{H2O}=2 Kb and X_{H2O}=0.7).
M, eutectic minimum; 830, 850, liquidus temperatures (°C).
b) Σ(La+Ce+Nd) plotted against the T°C estimated from the Zircon thermometry (Watson and Harrison, 1983) for the Bantoum granitoids. Symbols as in Fig. 5.

(Nabelek *et al.*, 1992) and Gangotri leucogranites (Scaillet *et al.*, 1990). Although the pressure of formation of these granites is not known, the high Or content, and therefore the generation under water-undersaturated conditions give constrains on the temperature of the melting reactions. Assuming a water activity of approximately 0.5, the minimum melt temperature is 775°C at 2 kb. This temperature would be slightly lower at higher pressure. However, the high Or content suggests that the melts leading to the crystallization of all granitic samples were formed at relatively high temperatures.

Many granitic magmas are saturated for numerous minor phases in certain trace elements, making possible their utilization as possible geothermometers in some determined circumstances. Saturation temperatures have been experimentally calibrated for zircon as a function of Zr content in granitic magma by Harrison and Watson (1983)

and Watson and Harrison (1983). The method supposes at first approximation that granitic magmas are close in composition to minimum liquid, and that the crystals of zircon are really magmatic (not inherited and free of any inherited core). Calculated temperature is that of the first appearance of zircon in the magma, and it could indicate either the temperature at which the melt was extracted from its source (Miller et al., 1988), or the temperatures marking the different stages of differentiation in case of fractional crystallization (e.g Scaillet et al., 1990). Hence, the more felsic granites are good candidates for calculating zircon saturation temperatures because these are commonly near minimum-temperature melts from which zircon was removed either as precipitated crystals or by fractionation of restite after partial melting. We applied that thermometer to Bantoum granitoids to get an idea about the minimum temperature of the melts (assuming that there is no residual zircon inherited from the source). Temperature ranges obtained are: 846-866°C (biotite-granites II), 812-839°C (biotitegranites I), 751-779°C (granitic dikes) 719-745°C (leucogranite) (Table 1). This thermometer can not be applied to the quartz-monzonites because these high Mg- and Ca-rocks were probably not derived from melting of crustal protoliths and the melts were not saturated with respect to zircon. Leucogranites have lower temperatures, which are consistent with those of crustal peraluminous granites; but biotite-granites have high temperature that are rather common in A-type granites (King et al., 2001). Chappell et al. (1998) subdivided I-type granite in two groups of low- and hightemperature granites. They assumed that high-temperature I-type granitoids are characterized by chemical evolutions testifying their origin from a mafic source followed by fractional crystallization; instead, those of low temperature show evidence of chemical evolutions testifying they are originated from a partial melting of quartzofeldspathic material followed by restite separation.

Figure 12b is the plot of the sum of LREE (La, Ce, Nd) against temperature calculated using zircon solubility method. It shows a good positive correlation and a discrimination between different granitic groups. It also suggests that concentration of Zr and LREE in these rocks may be controlled by the crystallization of zircon and other accessories phases (allanite, monazite) that used to concentrate LREE. This result is consistent with observations previously made by Broska *et al.* (2000) or Scaillet *et al.* (1990).

4.6. Sr-Nd isotopes results

Whole-rock Nd and Sr isotopic analyses were carried out on 18 samples of various petrographic groups from the Bantoum area, including quartz-monzonites, biotite-granites I, granitic dike and leucogranite. Results are listed in Table 2. Sm-Nd bulk-rock system in all of the cases couldn't define a meaningful isochron age because of narrow variation range in Sm/Nd ratios (Table 2).

Four samples of quartz-monzonites show an Rb-Sr whole-rock isochron age of 742 \pm 61 Ma, with an initial Sr isotopic ratio of 0.70681 \pm 0.00051 and a MSWD of 0.22 (Fig. 13). Their model ages calculated relative to depleted mantle (T_{DM}) range from 1.9 Ga to 2.1 Ga.

The biotite-granites I display T_{DM} ranging from 2.1 to 2.3 Ga for most of samples (sample S4.5b gives a T_{DM} =2.7 Ga) and biotite-granites II sample S3.3b has a T_{DM} =1.7 Ga.

Table 2. Rb-Sr and Sm-Nd isotopic analytical data for gneisses and granitoids from the Bantoum area. Number in parenthesis is the 20 error in the two last digits of the atomic ratio.

| (ppm) (ppm) 170 Biotite-granite I S4.5b 104 19.3 0.511; S4.3 122 16.9 0.511; S2.5 81.8 11.7 0.511; S2.3 129 21.2 0.511; B00 49.3 8.88 0.511; B14 40.1 8.81 0.511; Granitic dike S1.2 0.511; S1.4 47.1 10.3 0.511; Cuecogranite B15 33.4 7.68 0.511; Guartz-monzonite B47 98.1 15.0 0.511; S4.7 98.8 15.7 0.511; S4.5 98.8 15.7 0.511; B4.6 94.1 15.0 0.511; S4.7 98.8 15.7 0.511; S4.7 98.8 15.7 0.511; | | 0.1120 | (620 Ma) | V. F. COO. | | | | | | | |
|---|--|--------|----------|------------|------|-------|-------|--------------|--------|----------|----------|
| 1 19.3 16.9 11.7 21.2 8.88 8.81 11.7 21.2 11.2 11.2 11.2 11.3 11.3 15.0 15.0 15.0 15.0 15.7 | 1331(13) 1302(12) 1389(13) 1438(13) 1695(14) | 0.1120 | | (620 Ma) | (Ga) | (mdd) | (mdd) | 31/ 31 | NO/ OI | (620 Ma) | (620 Ma) |
| 19.3 16.9 11.7 21.2 8.88 6 8.81 II 21.2 1 10.3 1 7.68 1 7.68 | 1331(13) 1302(12) 1389(13) 1438(13) | 0.1120 | | | | | | | | | |
| 16.9 11.7 21.2 21.2 8 8.88 6 8.81 II 21.2 1 10.3 1 7.68 II 15.0 | 1302(12) 1389(13) 1438(13) 1695(14) | 0.1120 | 0.510876 | -18.8 | 2.7 | 224 | 266 | 0.735405(11) | 2.45 | 0.71378 | 142 |
| 11.7 21.2 3 8.88 6 8.81 II 21.2 1 10.3 4 7.68 1 15.0 | 1389(13) 1438(13) 1695(14) | 0.0839 | 0.510961 | -17.2 | 2.2 | 231 | 205 | 0.742631(12) | 3.28 | 0.71360 | 140 |
| 21.2 8 8.88 6 8.81 II 21.2 1 10.3 1 7.68 1 15.0 | 1438(13) 1695(14) | 0.0868 | 0.511036 | -15.7 | 2.1 | 199 | 182 | 0.740011(12) | 3.17 | 0.71195 | 116 |
| 8.88 6 8.81 II 21.2 1 10.3 1 7.68 1 15.0 1 15.0 | 1695(14) | 0.0995 | 0.511034 | -15.7 | 2.3 | 211 | 175 | 0.742167(13) | 3.50 | 0.71124 | 106 |
| 6 8.81 II 21.2 [1 10.3] 1 7.68 II 15.0 8 15.7 | | 0.1090 | 0.511252 | -11.5 | 2.1 | 305 | 121 | ı | 1 | | |
| 11 21.2 1 10.3 1 7.68 1 15.0 1 15.0 | 0.511414(12) | 0.0887 | 0.511054 | -15.3 | 2.1 | 217 | 200 | 0.739163(11) | 3.14 | 0.71139 | 108 |
| 21.2 1 10.3 1 7.68 1 15.0 1 15.0 | | | | | | | | | | | |
| 1 10.3 4 7.68 nite 15.0 8 15.7 | 0.511902(12) | 0.1038 | 0.511480 | -7.00 | 1.7 | 153 | 27.8 | 1 | , | | |
| 1 10.3 4 7.68 mite 15.0 8 15.7 | | | | | | | | | | | |
| 1 7.68 nite 15.0 8 15.7 | 0.511700(13) | 0.1320 | 0.511163 | -13.2 | 2.7 | 257 | 97.3 | 0.788235(12) | 7.71 | 0.72005 | 231 |
| 7.68 | | | | | | | | | | | |
| 15.0 | 0.511741(12) | 0.1390 | 0.511176 | -13.0 | 2.9 | 370 | 45.5 | 0.928388(12) | 24.0 | 0.71584 | 171 |
| 94.1 15.0 98.8 15.7 | | | | | | | | | | | |
| 98.8 | 0.511663(13) | 0.0962 | 0.511272 | -11.1 | 1.9 | 171 | 1065 | 0.711757(14) | 0.466 | 0.70764 | 55 |
| | 0.511664(11) | 0.0959 | 0.511274 | -11.0 | 1.9 | 171 | 1042 | 0.711821(14) | 0.476 | 0.70761 | 55 |
| S3.7c 79.4 12.1 0.511 | 0.511530(13) | 0.0919 | 0.511157 | -13.3 | 2.0 | 163 | 609 | 0.715224(12) | 9/1/0 | 0.70836 | 65 |
| S3.6 67.0 10.8 0.511 | 0.511565(11) | 0.0977 | 0.511168 | -13.1 | 2.1 | 192 | 615 | 0.716180(14) | 0.904 | 0.70819 | 63 |
| Hornblende-biotite gneiss | | | | | | | | | | | |
| S3.11 139 30.1 0.511 | 0.511943(12) | 0.1304 | 0.511413 | -8.32 | 2.2 | 139 | 57.7 | 0.852550(13) | 7.09 | 0.78988 | 1224 |
| S3.12 143 33.3 0.511 | 0.511953(14) | 0.1403 | 0.511383 | -8.90 | 2.5 | 158 | 39.5 | 0.937834(19) | 11.8 | 0.83311 | 1838 |
| B25 99 20.4 0.511 | 0.511802(13) | 0.1244 | 0.511296 | -10.6 | 2.3 | 205 | 91.8 | 0.858368(24) | 6.55 | 0.80048 | 1374 |
| B31 | | · | ı | 1 | 1 | 145 | 90.1 | 0.804744(13) | 4.70 | 0.76314 | 844 |
| Plagioclase-rich gneiss | | | | | | | | | | | |
| B43 43.8 10.1 0.512 | 0.512278(12) | 0.1393 | 0.511712 | -2.48 | 1.8 | 74.0 | 306 | 0.716931(09) | 0.699 | 0.71075 | 66 |

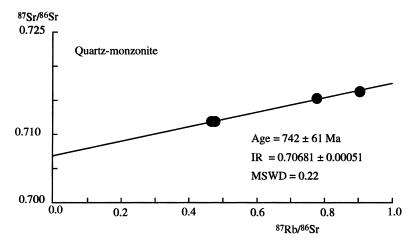


Fig. 13. Rb-Sr isochron diagram for the Bantoum quartz-monzonites.

For the granitic dike sample S1.4 T_{DM} =2.7 Ga. The leucogranite sample B15 shows an oldest T_{DM} of 2.9 Ga. In the Gneisses, T_{DM} range from 2.5 Ga-2.2 Ga to 1.8 Ga (plagioclase-rich gneiss B43).

5. Discussion and conclusion

The Rb-Sr whole-rock isochron age obtained for the quartz-monzonites is older than different Rb-Sr ages ranging from (575 to 657 Ma) obtained on other plutonic complexes in west Cameroon (Kwékam, 1993; Talla, 1995; Tagné Kamga, 1994); the neighboring formations of the Bangangte syenitic complex are dated at 603–606 Ma (Rb-Sr whole rock isochron; Tchouankoué, 1992), and a metadiorite at Bangoua (30 km west of Bantoum) is dated 628 Ma (U-Pb zircon; Toteu *et al.*, 2001). These ages are interpreted to be intrusion or crystallization ages.

In order to test a possibly mixing in the source materials of granitoids, we used the model initial ⁸⁷Sr/⁸⁶Sr ratios vs 1/Sr variation diagrams. For the ages above 680 Ma, plots of quartz-monzonites are scattered and biotite-granites show a meaningless strong negative correlation; and moreover, leucogranite gives an unrealistic initial 87Sr/86Sr ratio that is lower than 0.696; then an age above 680 Ma couldn't be applied the leucogranites. For 620 Ma (Fig. 14), quartz-monzonites show a positive correlation that could be interpreted as a mixing trend. We then assume that the 742 Ma isochron age to be rather a mixing line. The ongoing Rb-Sr and Sm-Nd mineral ages being carried out on selected samples from study area will help us ascertaining that issue. Calculated age for 5 samples of biotite-granites gives $543\pm173\,\mathrm{Ma}$ but with poor precision due to narrow range in variation of ⁸⁷Rb/⁸⁶Sr, and ⁸⁷Sr/⁸⁶Sr=0.71588± 0.00753. Despite of this large uncertainty, this result is within the range of ages of Pan-African intrusives in CAFB, and it confirms at least that biotite-granites are younger than quartz-monzonites as it is suggested by field relationships (e.g. no enclave observed in quartz-monzonites, but both enclaves of quartz-monzonites and gneisses are found in biotite-granites).

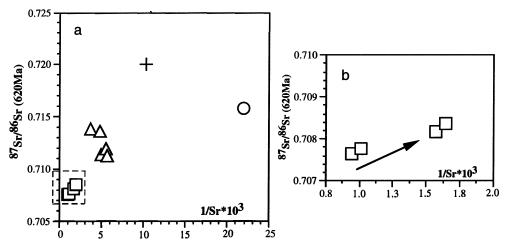


Fig. 14. a) 87Sr/86Sr (620 Ma) vs 1/Sr variation diagram for the granitoids. b) Enlargement of the squared area in a) for quartz-monzonites; the arrow shows the possibly mixing trend. Symbols as in Fig. 5.

Extrapolations to $T_{\rm DM}$ for plutonic rocks can have large uncertainties due to possible REE fractionation during magma genesis if they are much younger than their source terranes, as is the case for ca 500–650 Ma Pan-African granitoids intruding and presumably derived from Archean to Paleoproterozoic basement in parts of the CAFB. As a result, we have chosen 620 Ma as the main reference time for Nd compositions in the past as well as Sr and Nd initial ratios, since we can extrapolate accurately to that time without worrying about possible fractionation of Sm-Nd during either magma genesis or high-grade metamorphism; and furthermore it corresponds to the average age of most of plutons in CAFB.

For quartz-monzonites, initial 87 Sr/ 86 Sr ratios (SrI) calculated using 620 Ma vary from 0.70761 to 0.70836; their ε_{Nd} (620 Ma) range from -13.3 to -11.0.

The biotite-granites I display model SrI (620 Ma) ranging from 0.71124 to 0.71378. They have model initial 143 Nd/ 144 Nd ratios of 0.51130–0.51170 and $\varepsilon_{\rm Nd}$ (620 Ma) from -18.8 to -11.5.

The granitic dike sample S1.4 gives a high initial Sr isotopic ratio (620 Ma) = 0.72005. It's model initial ¹⁴³Nd/¹⁴⁴Nd ratio = 0.51116 and the ε_{Nd} (620 Ma) is -13.2.

The leucogranite sample B15 has a SrI (620 Ma) = 0.71584. It's model initial 143 Nd/ 144 Nd ratio = 0.51118 and the ε_{Nd} (620 Ma) is -13.0.

Gneisses have higher model SrI, whose values show a larger spread. These model SrI (620 Ma) are 0.71075 (B43, Pl-rich gneiss), 0.76314 (B31, Grt-gneiss) and 0.78988–0.83311 (Hb-Bt gneiss). Model initial 143 Nd/ 144 Nd ratios are 0.51138–0.51171, with $\varepsilon_{\rm Nd}$ (620 Ma) ranging from $-10.6^{\sim}-8.3$ to -2.5 for the plagioclase-rich gneiss sample B43.

The variation in the values of the initial ⁸⁷Sr/⁸⁶Sr ratios in the different groups of granitoids is indicative of the heterogeneity in their source rocks. These ⁸⁷Sr/⁸⁶Sr initial ratios are high suggesting parental magmas were, therefore, derived from, or contaminated with different crustal sources with relatively high concentration of radiogenic

⁸⁷Sr; the lowest values observed in quartz-monzonites show that they have more deep-seated origin.

Overall, T_{DM} ages of investigated samples range from 1.8 to 2.9 Ga with most of them being grouped around 2.1-2.3 Ga, and they have large negative values of $\varepsilon_{\rm Nd}$ (620 Ma) from -19 to -13. The large negative initial ε_{Nd} values of the granitic rocks is evocative of their long and complex crustal history. These Nd isotopic signatures confirm the presence of remnants of a lower Paleoproterozoic or Archean crust in that domain of the CAFB (Penaye et al., 1993; Toteu et al., 1994a, 2001). Relatively lower values of initial Sr and Nd ratios observed in quartz-monzonites point to an involvement of a significant amount of juvenile magma. This result is in agreement with observations by Toteu et al. (2001) that Pan-African granitoids southeast of the Banyo-Tcholliré shear zone (Fig. 1) are generated mainly by melting of older (Eburnean) crust with limited inputs of juvenile magmas. Sr-Nd isotopic characteristics of gneisses are significantly different from those of associated granitoids as illustrated in Fig. 15a, showing that they may have different history and no genetic link. For the plagioclaserich gneiss sample B43, higher ε_{Nd} value (-2.5) and lowest T_{DM} age suggests it may contain volcanic material in its protolith. The field of compositions of some Archean syenites and high-K granitoids from the Ntem complex is shown on Fig. 15b. The Bantoum granitoids plot in the continuation of the field of these Archean rocks, suggesting that source materials similar to these high-K granitoids and syenites from the Archean Ntem complex (Tchameni et al., 2000, 2001) could be good candidate for the protoliths of the Bantoum granitoids. On the ε_{Nd} vs time diagram (Fig. 16), the evolution of ε_{Nd} values of Bantoum granitoids emphasizes that they can be formed by partial melting of old material with an isotope signatures similar to those of Archean crust from of the Ntem complex during the Pan-African orogenesis. However, combined zircon U-Pb ages and Sm-Nd data in other parts of CAFB in Cameroon have led

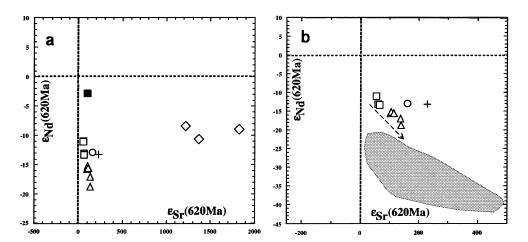


Fig. 15. a) Model initial ε_{Nd} vs ε_{Sr} diagram for granitoids and gneisses; b) Model initial ε_{Nd} vs ε_{Sr} diagram for the Bantoum granitoids. Shaded area corresponds to compositions of Archean syenites and high-K granitoids from the Congo craton (data from Tchameni et al., 2000; 2001). Symbols as in Fig. 5.

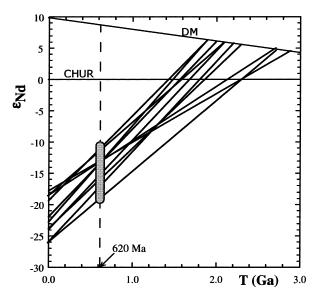


Fig. 16. $\varepsilon_{\rm Nd}$ vs time variation diagram for granitoids, showing the significant contribution of a Paleoproterozoic to Archean crust in their protolith.

Toteu et al. (1994a, 2001) to conclude that the Archean crust was not directly involved as the source for protoliths of Pan-African granitoids as it was completely recycled during the 2100 Ma (Eburnean) event.

The overall geochemical features of investigated rocks are compatible with compositions of calc-alkaline magmas in orogenic domains and are similar to those documented in numerous huge Pan-African plutons in west Cameroon (e.g. Nguiessi Tchankam et al., 1997); particularly their hyperpotassic calc-alkaline trends are characteristic of a continental collision framework (Black and Liegeois, 1993; Liegeois et al., 1994). Therefore within-plate granite (WPG) trace element character of granites II and gneisses (Fig. 10) is probably not indicative of the tectonic framework during their formation, and could be interpreted as being inherited from the source of their protolith. Liegeois et al. (1994) attributed such trace element character of granitoids from the Tuareg Shield to crustal delamination operating in a syn- to post-collisional geodynamic setting.

Our data also suggest that the difference in geochemical nature between the different groups of granitoids in the study area is not necessarily related to changes in tectonic framework but is more probably due to variations in the mechanisms of magma production in the crust. Besides the source material, water contents and melting temperatures for example are known to be critical variables as evidenced by investigations of Clemens and Vielzeuf (1987) and Johannes and Holtz (1990). Low water activities or melting due to fluid-absent dehydration reactions explain partial melting at higher temperatures for the formation of the granites. In such a case, there is more efficient dissolution of residual mineral phases such as zircon, apatite, monazite and sphene in the melts, and as a result, an increase in the HFSE and REE contents in the

produced melts. Examples of plutonism in which, depletion in water of the crustal source have imposed changes in the melting conditions during subsequent events of production of granite magmas, have been documented by Whalen *et al.* (1987) and Pimentel *et al.* (1996). In most of the cases, it is likely that the input of heat required to promote large-scale melting of the crust was most probably provided by the introduction of mafic magmas into the crust (Roberts and Clemens, 1993).

Neither geochemical features nor isotopic data of the Bantoum area granitoids supports the eventuality of a formation of biotite-granites by differentiation processes from quartz-monzonites. On the basis of these chemical and isotopic characteristics, it appears that the investigated rocks in the Bantoum area could be subdivided at least into four groups: quartz-monzonites, biotite-granites, leucogranites and gneisses. Each group among granitoids crystallized at different temperatures and from different source materials during the same orogenic event; but there is an important participation of the older Eburnean material in their protoliths. Among biotite-granites, a few part (biotite-granites II) share the same geochemical features as the hornblende-biotite gneisses, notably similar chondrite-normalized trace elements and REE distribution patterns and their position in the same field in tectonic discrimination diagrams, suggesting the contribution of these gneisses to the protolith of some biotite-granites. Results of melting and crystallization experiments carried out on an amphibole-biotite gneiss (Nzolang et al., in prep.) show that compositions of experimental liquids produced from that gneiss are comparable solely to those of granites II, supporting that amphibolebiotite gneiss is a possible source for biotite-granites II.

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